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L'energia chimica muove la vita

ATTI DEL CONGRESSO

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Olefin metathesis has emerged as a powerful tool in both organic and polymer chemistry. A variety of applications has become accessible, including ring closing metathesis (RcM), cross metathesis (RCM), acyclic diene metathesis polymerization (ROMP). OleMET) and ring opening metathesis polymerization (ROMP). Diverse ruthenium-based complexes can catalyze these transformations. Among those systems, N-heterocyclic (NHC) Ru complexes have emerged as the most versatile and efficient catalysts.

Catalytic performances of new Ru-based complexes bearing saturated chiral NHC ligands in some standard RCM, CM and ROMP reaction are examined

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## INO-PO-30

Application of the Shvo Catalyst in Bio-oil Upgrading. Hydrogenation of Fast Pyrolysis Oil and Related Model Compounds

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In the development of renewable energy carriers an attractive biofuel is the bio-oil obtained from flash pyrolysis of lignocellulosic biomass. The crude bio-oil (consisting of acids, aldehydes, ketones, up to 30% water and polymeric materials)2 requires upgrading to improve its caloric value and storage stability. Hydrotreatment using classical heterogeneous catalysts or zeolites increases the caloric value of upgraded bio-oil. Acting in the same way recent developments show that homogeneous catalysts could be attractive alternatives as these operate at milder conditions.3 Here we report on application of the Shvo catalyst  $[\{[3,4-(4-MeO-C_0H_a),-2,5-Ph,(\eta^5-C_1CO)],H\}Ru,(CO),(\mu-H)],$  an efficient homogeneous hydrogenation catalyst, in the reduction of polar double bonds in model mixture (vanillin, cinnamaldehyde, 4-methyl-acetophenone, glycolaldehyde, acetol, acetic acid). The catalyst works with high conversions in acidic aqueous and organic model mixtures. Preliminary results obtained on a real matrix (bio-oil from poplar), show the reduction of polar double bonds, with increased stability.

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INO-PO-31

## CuO/Ce<sub>(1-x)</sub>Zr<sub>x</sub>O<sub>2</sub> porous layered materials for the CO Preferential Oxidation

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Layered mixed oxides, consisting in a structure of stacked sheets, are very a structure of stacked sheets, are very design, controlled accessibility to the active sites and high surface area, and also show

an interesting behaviour as catalysts. In this work, the preparation of a series of COVCe<sub>1</sub>, 27, 0, poursu layered materials is described. The samples were characterized by X-ray product diffraction (XPD), thermogravimetry (TGA-DTG), SEM microscopy, N, physisorption at 77 K, temperature-programmed reduction (HL-TPB) and X-ray photoelectron spectroscopy (XPS). The activity of these layered materials was evaluated in the preferential voidation of CO (CO-PROX)<sup>2</sup> in hydropen-rich gas stream (1.2% CO, 1.2% O, .9% H, O.15% CO, 0.-10% H,O, He balance) and their catalytic performance was compared with that of a CuO-CoO, reference sample. Correlations between catalytic activity and physics-chemical properties of the materials were mixth at the catalytic performance of the materials were mixth and only one common the materials were mixth.

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## INO-PO-32

Photocatalytic Degradation of 4-Nitrophenol by Pyrazinoporphyrazines Deposited on TiO, (anatase) Maria Pia Donzello, Claudio Ercolani, Fabrizio Monacelli, Giuliano Moretti, Piera More

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The porphyrazine macrocycle tetrakis-2,3-[5,6-di(2-pyridyl)pyrazino|porphyrazine (abbr. as [Py,1PyzPzH],) and its metal complexes [Py,1PyzPzH] (M = Zn°, Cu°, Pd°)-2 (Figuro) adsorbed on TiO, (anatase) particles (Ø = 170 mm) were proved able to significantly increment in the aqueous suspension the activity of pure TiO, in the

photodegradation of 4-nitrophenol. The efficiency of the combined action macrocycle/TiO<sub>2</sub>, higher than other studied related tetrapyrrolic analogs, depends on the type of the porphyrazine core and follows the order: [LCu]/TiO<sub>2</sub> > [LZn]/TiO<sub>2</sub> > [LPd]/TiO<sub>2</sub> > TiO<sub>3</sub> (LP, Py, Py, Pz)<sup>2</sup> = [LPd]/TiO<sub>3</sub> > TiO<sub>4</sub> > [LPd]/TiO<sub>5</sub> > TiO<sub>5</sub> (LPd)/TiO<sub>5</sub> > TiO<sub>5</sub> = [LPd]/TiO<sub>5</sub> > TiO<sub>5</sub> > [LPd]/TiO<sub>5</sub> > [LPd]/TiO<sub>5</sub>

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