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ANALYSIS OF MAJOR AIR POLLUTANTS AND SUBMICRON PARTICLES IN NEW YORK CITY AND LONG ISLAND

M. Masiol, ¹ P. K. Hopke, ¹ H.D. Felton, ² B.P. Frank, ² and O.V. Rattigan²

 ¹ Center for Air Resources Engineering and Science, Clarkson University, Potsdam, NY 13699
 ² NYS Department of Environmental Conservation, Division of Air Resources, BAQS, Albany NY 12233-3256

HIGHLIGHTS

- > One-year air quality data measured hourly at 2 urban sites in NYC and Long Island
- > Particle number resolved over 6 bins, gases, PM2.5 mass, EC/OC and sulfate measured
- > Primary and secondary organic carbon are estimated using the EC tracer method
- > Seasonal, weekly, diurnal cycles and relationships with weather are investigated
- > Location of the potential sources and role of regional transport are discussed

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Analysis of Major Air Pollutants and Submicron Particles 1 in New York City and Long Island 2 3 M. Masiol, P. K. Hopke, H.D. Felton, B.P. Frank, and O.V. Rattigan 4 5 ¹ Center for Air Resources Engineering and Science, Clarkson University, Potsdam, NY 13699 6 ² NYS Department of Environmental Conservation, Division of Air Resources, BAQS, Albany 7 8 NY 12233-3256 9 10 **Abstract** A year-long sampling campaign of major air pollutants and submicron particle number size 11 12 distributions was conducted at two sites taken as representative of city-wide air quality in New York City and Long Island, respectively. A number of species were quantified with hourly time 13 14 resolution, including particle number concentrations in 6 size ranges (20-30 nm, 30-50 nm, 50-70 15 nm, 70-100 nm, 100-200 nm, and >200 nm), nitrogen oxides, sulfur dioxide, ozone, carbon 16 monoxide, methane, non-methane hydrocarbons, PM_{2.5} mass concentration and some PM major 17 components (sulfate, organic and elemental carbon). Hourly concentrations of primary and 18 secondary organic carbon were estimated using the EC tracer method. Data were matched with 19 weather parameters and air parcel back-trajectories. A series of tools were thus applied to: (i) 20 study the seasonal, weekly, diurnal cycles of pollutants; (ii) investigate the relationships amongst 21 pollutants through correlation and lagged correlation analyses; (iii) depict the role of atmospheric 22 photochemical processes; (iv) examine the location of the potential sources by mean of 23 conditional bivariate probability function analysis and (v) investigate the role of regional 24 transport of air masses to the concentrations of analyzed species. Results indicate that 25 concentrations of NO_x, SO₂, CO, non-methane hydrocarbons, primary OC and EC are 26 predominantly determined by local sources, but are also affected by regional transports of

By providing direct information on the concentrations and trends of key pollutants and submicron particle number concentrations, this study finally enables some general considerations about air quality status and atmospheric processes over the New York City metropolitan area.

polluted air masses. On the contrary, the transport of continental polluted air masses has a main

effect in raising the concentrations of secondary PM_{2.5} (sulfate and secondary organic carbon).

Keywords: Air quality; New York City, Nitrogen oxides; Organic carbon; Sulfate; Particle size.

1. INTRODUCTION

Over the last several decades, an increasing number of epidemiologic studies have concluded that ambient air pollution and particularly airborne particulate matter (PM) are causal risk factors for many adverse human health effects, including respiratory illnesses (Götschi et al., 2008; Laumbach and Kipen, 2012; Adam et al., 2015), cardiovascular diseases (Anderson et al., 2012; Shah et al., 2013; Franklin et al., 2015) and carcinogenic effects (Turner et al., 2011; Pope et al., 2011; Loomis et al., 2013). Given the biological evidence, outdoor air pollution and airborne particulate matter (PM) have been categorized in class 1 by the International Agency for Research on Cancer in 2013, i.e. known carcinogenic to human beings (Straif et al., 2013).

A significant drop in the ambient concentrations of many air pollutants has been measured in many developed countries (including U.S.) as a result of implementation of legislation and regulations through the application of successful abatement technologies and other mitigation measures (Parrish et al., 2011; Colette et al, 2011). Current United States air quality is regulated through the National Ambient Air Quality Standards (NAAQS) that set limit values to be attained across U.S. for pollutants considered harmful to public health and the environment, i.e. nitrogen dioxide (NO₂), ozone (O₃), sulfur dioxide (SO₂), carbon monoxide (CO), PM with aerodynamic diameters less than 10 and 2.5 μm (PM₁₀ and PM_{2.5}, respectively) and particle-bound lead.

New York City (NYC) is the most populous city in the U.S. with ~8.5 million inhabitants in 2014. However, the population reaches almost 20 million inhabitants when considering the whole New York–New Jersey–Pennsylvania metropolitan area. Air quality has generally improved since 1970s across NYC (NYC Department of Health and Mental Hygiene, 2013a). For example, Parrish et al. (2011) have reported slow improvements for both ozone and PM_{2.5} between 2000 and 2010, while Duncan et al. (2016) have observed a strong decrease (-45.5%) in NO₂ from 2005 to 2014. Downward trends in air pollutants across New York were observed by Buckley and Mitchell (2011), while substantial decreases in PM_{2.5} mass and major components across New York since the early 2000s were reported by Rattigan et al. (2016). Currently, NAAQS are generally not exceeded in NYC (NY State Dep. Environmental Conservation, 2016), except for ozone (marginal to moderate non-attainment). However, air pollution still remains a serious concern for policy-makers and the scientific community in NYC as well as in

the whole metropolitan area. Even with attainment of the NAAQS, it has been estimated that ambient concentrations of PM_{2.5} still contribute to more than 3000 deaths every year, 2000 hospital admissions for lung and heart conditions, and approximately 6000 emergency department visits for asthma in children and adults (NYC Department of Health and Mental Hygiene, 2013a;b).

Despite the number of stations measuring air quality and the large number of scientific studies available in literature, limited data are still available for some air pollutants, including submicron particles (SMPs, <1 µm) and ultrafine particles (UFPs, less than 100 nm in diameter), which have been widely associated with severe adverse effects upon human health (e.g., Knibbs et al., 2011; Strak et al., 2012; Ostro et al., 2015; Lanzinger et al., 2016). Moreover, limited studies have focused on air pollution climate, i.e. on the relationships amongst air pollutants and weather.

This study investigates air pollutions in NYC and Long Island through the analysis of hourly-resolved concentrations of common gaseous air pollutants, PM_{2.5} mass and PM_{2.5}-bound major species, submicron particle number concentrations and their size distributions, with special emphasis for UFPs. Intensive sampling campaigns were simultaneously conducted at two sites: Queens (NYC) and Eisenhower Park (Long Island). A large set of air pollutants was measured over 1 year (2009-2010) at 1 h intervals. Datasets include: (i) key gaseous air pollutants, i.e. nitrogen oxides (NO+NO₂ \equiv NO_x), sulfur dioxide (SO₂), ozone (O₃), carbon monoxide (CO), methane (CH₄), non-methane hydrocarbons (NMHCs), total hydrocarbons (THCs); (ii) PM_{2.5} mass and PM_{2.5}-bound species, such as sulfate (SO₄²-), organic (OC), elemental (EC) and total (TC) carbon; (iii) particle number concentrations (PNC) over 6 size ranges (20-30 nm, 30-50 nm, 50-70 nm- 70-100 nm, 100-200 nm, and 200 to \sim 1000 nm). The seasonal, weekly, and diurnal cycles of the air pollutants were investigated in detail along with several derived variables. The concentrations of primary and secondary organic carbon were estimated using an EC tracer method. Data were matched with micro-meteorological parameters and analyzed to explore the possible effects of weather upon the air quality and to locate the probable local emission sources. A back-trajectory-based hybrid model was applied for assessing the potential role of regional and long-range transports upon air quality. These results give important insights into pollutant sources, influences upon concentrations and key inter-relationships in pollutant behaviors and

presents a synthesized description of pollution climate regimes for typical urban sites in NYC or even for large cities across northeastern U.S.

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2. MATERIALS AND METHODS

2.1 Site description

- Measurements were made from June 2009 to July 2010 at two sites categorized as urban, i.e.
- 100 representative of city-wide background pollution concentrations. Queens College (QC: 40°
- 44.225'N; 73° 49.295'W; 24 m a.s.l.) is located in a high population density section of Queens
- 102 County (Figure 1) and it is representative of the New York City metropolitan area. QC is also
- affected by emissions from two traffic highways, i.e. the Long Island Expressway (I-495) and the
- 104 Van Wyck Expressway (I-678). Eisenhower Park (EP: 40° 44.586'N 73° 35.132'W; 25 m a.s.l.)
- is situated on the border of a 3.8 km²-wide public park located in Nassau County. The park is
- devoted to many entertainment and sport activities and is backed by unoccupied parklands.
- However, EP is also affected by: (i) heavy road traffic as it is adjacent to a 4 lane roadway that
- runs parallel to the site in a NNW direction, within 22 m of a roadway in a W-WSW direction,
- and within 53 m of the roadway in a S direction; and (ii) emissions from a waste-to-energy
- 110 conversion plant and from several restaurants close to the site (Figure 1).

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2.2 Experimental

- Instruments were operated continuously over the sampling period to quantify ambient air pollutant concentrations with 1 h resolution time. A more extensive range of pollutants were
- air pollutant concentrations with 1 h resolution time. A more extensive range of pollutants were
- monitored at QC (NO, NO₂, NO_x, SO₂, O₃, CO, CH₄, NMHCs, THC, PM_{2.5}, SO₄²⁻, OC, EC, opt.
- 116 EC, TC and SMP) than at EP (NO, NO₂, NO_x, SO₂, PM_{2.5} and SMP).
- 117 Two ultrafine particle monitors (TSI Model 3031) were deployed at both sites to
- measure particle size distributions from 20 nm (electrical mobility diameter) to approximately
- PM_{2.5} μ m in six channels: 20-30 nm (ch1), 30-50 nm (ch2), 50-70 nm (ch3), 70-100 nm (ch4),
- 120 100-200 nm (ch5) and >200 nm (ch6). The instruments operated with PM_{2.5} cyclones, i.e. the
- 121 upper size was set to particles with an aerodynamic diameter less than 2.5 μm. The operating
- principle of the UFP monitor is based diffusion charging of particles using a "Corona-Jet"

charger (Medved et al., 2000), followed by size segregation in a differential mobility analyzer and detection via a sensitive electrometer.

U.S. EPA equivalent or reference methods were used for the gas measurements (Thermo Electron Corporation model 42C NO-NO₂-NO_x; model 43C SO₂; model 49C O₃; model 48C CO). Methane, Non-Methane Hydrocarbons (NMHCs), and total hydrocarbons (THC) were analyzed using flame ionization detector method (model APHA-360, HORIBA Ltd., Japan).

PM_{2.5} mass concentrations were measured with tapered element oscillating microbalances (Thermo-Fisher Scientific TEOM 1400ab) with 2.5 μm cut size inlets. However, a TEOM-FDMS system was deployed at QC, while a simple TEOM was used at EP. The main concern with the use of the simple TEOM technique is the loss of the more volatile component (principally some semi-volatile hydrocarbons and nitrates) because the inlet is held at a temperature ~50°C. A non-linear Julian Day based adjustment was thus applied to the TEOM data: it removes much of the seasonal bias and gives approximate comparability with the Federal Reference Method (FRM). Details are reported in Felton (2005).

Particulate sulfate was determined with a sulfate analyzer (Thermo Scientific model 5020i), while particulate carbonaceous components (OC, EC, TC) were quantified by a semi-continuous carbon aerosol analyzer (Sunset Lab., Tigard, OR, USA) operating a modified NIOSH 5040 thermo-optical protocol on an hourly sample collection similar to Bauer et al. (2009). For EC, both the thermal (EC) and "black" optical EC (opt. EC at 632 nm) were quantified.

2.3 Weather data and meteorology-based models

Ambient air temperature, wind speed and direction, relative humidity (RH) and atmospheric pressure were recorded hourly at weather stations which were chosen as representative of the average weather conditions across the study area: LaGuardia and Republic airports, respectively, for QC and EP. Since no direct solar radiation (SR) data were available at the weather stations, the total direct and diffuse SR (METSTAT-modeled) was retrieved from the National Solar Radiation Database (NREL, 2012).

Conditional probability function (CPF) analysis is widely used to locate the potential local sources affecting a site. It is performed by matching wind direction data with air pollutants concentrations (Kim et al., 2003; Pekney et al., 2006; Kara et al., 2015). More recently, this

approach was extended to the bivariate case to produce a conditional bivariate probability function (CBPF) plots using wind speed as a third variable plotted on the radial axis: details of the method are provided in Uria-Tellaetxe and Carslaw (2014). Essentially, CBPF assesses the probability that a source contribution from a given wind direction/speed sector exceeds a predetermined threshold criterion (concentrations >75th percentile, in this case). This approach was used to identify the most probable local emission sources.

Back-trajectories were computed by the NOAA/ARL hybrid single particle Lagrangian integrated trajectory (HYSPLIT4) model (Stein et al., 2015; Rolph, 2016) from the NCEP/NCAR Reanalysis data. The HYSPLIT parameters were set as -96 h run time with a starting height of 500 m a.g.l. The potential source contribution function (PSCF) (Ashbaugh et al., 1985; Lee and Hopke, 2006; Hopke, 2016) was used to identify regional scale source locations. PSCF calculates the conditional probability that a source is located at latitude *i* and longitude *j* as:

$$167 PSCF = \frac{m_{i,j}}{n_{i,j}}, (1)$$

where $n_{i,j}$ is the number of times the trajectory endpoints fell into the i,j-th cell in the domain grid, and $m_{i,j}$ represents the number of times the concentrations exceed a threshold (concentrations >75th percentile, in this case). The cells with high PSCF probabilities indicate a likely source location. Since cells with few data may bias the results, a weighting factor is applied to reduce their effect. The details are provided in Polissar et al. (2001).

Data were analyzed using R version 3.2.2 (R Development Core Team, 2008) and a series of packages including 'openair' (Carslaw and Ropkins, 2012). Details of polar plot analysis shown in this study are given in Carslaw et al. (2006) and Uria-Tellaetxe and Carslaw (2014).

3 RESULTS

Annual concentrations of measured pollutants at the two sites (all available data recorded during the whole study period) are given as box and whisker plots (Figure 2). On average, results show that gaseous species measured at both sites have similar concentrations, following the order (in ppb) NO_x (28.9 and 24.5 in QC and EP, respectively)> NO₂ (19.6, 14.3)> NO (9.4, 10.3)> SO₂

(2.8, 2.3). Average PM_{2.5} mass concentrations were 10 and 9.4 µg m⁻³ at QC and EP, 184 respectively, while total SMPs averaged 8209 and 7596 particles cm⁻³. The size distributions 185 186 show the highest concentrations in the 30-50 nm range at both sites, followed by 20-30 nm. 187 Particles >200 nm had the lowest concentrations. QC experiences higher concentrations for all of 188 the measured air pollutants, except for NO. 189 At QC, annual PM_{2.5} mass concentrations are dominated by sulfate (20.9%) and OC 190 (20.6%), with EC accounting for 6.5%. Assuming the sulfate is fully neutralized by ammonia 191 and a OM/OC ratio in the range of 1.9–2.1, as reported for atmospheric aerosols sampled at the 192 Pittsburgh supersite (Polidori et al., 2008), ammonium sulfate accounts for ~25% of total PM_{2.5} mass $(2.7 \,\mu g \, m^{-3})$, while organic matter represents 39-43% (4-5 $\mu g \, m^{-3}$). 193 194 Additional variables were calculated and the results are shown in Figure 2. The 195 NO₂/NO_x ratio indicates the complex photochemistry in the NO-NO₂-O₃ system. NOx emissions 196 from combustion is primarily NO (~90%) and rapidly reacts with O₃ and radicals to form NO₂ 197 (Jenkin and Hayman, 2000; Finlayson Pitts and Pitts, 2000; Seinfeld and Pandis, 2006). Annual 198 average NO₂/NO_x ratios were found around 0.8 (0.82 at QC and 0.77 at EP). Total oxidant 199 concentration (OX=O₃+NO₂) was calculated as an indirect estimation of the atmospheric 200 oxidative potential (Kley et al., 1999). At QC, annual average OX concentrations were 43 ppb, 201 of which ozone contributed for 54% and NO₂ for the remaining 46%. The sulfate/SO₂ ratio may 202 give important insights upon the degree of SO₂ oxidation and on the consequent speciation of 203 sulfur into the two species (Stockwell and Calvert, 1983; Stockwell, 1994). Sulfate accounts for 204 30–60% of monthly average PM_{2.5} mass in the East U.S. (Hand et al., 2012a;b). The sulfate/SO₂ 205 ratio was calculated over the mass concentrations normalized for their sulfur content: i.e. converting SO₂ to µg m⁻³ and dividing the species by the respective molecular weights. Annual 206 207 average of sulfate/SO₂ ratio at QC was 0.4. This value indicates that the atmospheric sulfur is

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3.1 Primary and secondary organic carbon

primarily oxidized S(IV).

Primary (OC_{pri}) and secondary (OC_{sec}) organic carbon aerosols are important bulk constituents of PM_{2.5}. Secondary OC is formed in the atmosphere from the oxidation of volatile organic compounds (VOCs) of both biogenic (e.g., Lane and Pandis, 2007; Brown et al., 2009) and anthropogenic (e.g., Ng et al., 2007; Henze et al., 2008; Donahue et al., 2009) origins. It was estimated that OC_{sec} accounts for about 35-77% of total OC across the continental U.S. (Yu et

al., 2004; Cabada et al., 2004a; Yu et al., 2007; de Gouw et al., 2008). In this study, a widely

- used EC tracer method is applied to estimate the partitioning between OC_{pri} and OC_{sec} using
- 218 hourly resolved OC and EC data collected at QC. Briefly, since EC is a primary pollutant, the
- adopted empiric method assumes that OC_{pri} is dependent from EC:

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$$OC_{pri} = EC \cdot \left[\frac{OC}{EC}\right]_{pri} + b$$
 (2)

- where OC/EC_{pri} is the ratio for the primary sources affecting the site and b includes the non-
- combustion contribution to OC_{pri} and sampling artifacts (Turpin and Huntzicker, 1995; Strader et
- al., 1999; Lim and Turpin, 2002; Cabada et al., 2004a). Subsequently, OC_{sec} can be finally
- assessed as:

$$225 OC_{\text{sec}} = OC - OC_{\text{pri}} (3)$$

- The limitations and the uncertainties associated with EC tracer methods have been noted (e.g.,
- Day et al., 2015) and are mainly related to OC/EC_{pri}. This basic method relies on the ability to
- isolate periods which are dominated by primary EC and OC, i.e. low photochemical secondary
- organic aerosol (SOA) production. It assumes that primary EC and OC are derived from similar
- sources and exhibit similar temporal patterns throughout the year. The (OC/EC)_{pri} value is rarely
- 231 measured experimentally and, thus, needs to be estimated. A scatterplot of OC versus EC
- concentrations for all samples and for samples selected by adopting criteria similarly to Cabada
- et al. (2004a) is shown in Figure SI1. A clear edge is found at OC/EC=1.07, which is interpreted
- as the OC/EC_{pri}, whereas no intercept is detected ($b\approx0$). Following this procedure, the average
- concentration of primary OC was 0.84 μg C m⁻³, accounting for 35% of total OC, while average
- OC_{sec} was 1.55 µg C m⁻³ (65%). Similar results were previously observed at a NYC Bronx site
- 237 (Rattigan et al., 2010).

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3.2 Inter-site correlations

Since the Anderson-Darling test for the composite hypothesis of normality applied to the data revealed that variables are generally not normally distributed at p<0.05, the nonparametric Kruskal-Wallis ANOVA by ranks was applied to test the significance of

differences between the two sites. Results indicate that only SO₂, PM_{2.5}, and SMP in the range

100-200 nm concentrations were significantly different at p<0.05. The correlation coefficients were calculated for the data over the whole sampling campaign (Table SII) to examine the intersite relationships among pollutants. Results show that many of the variables measured at both sites have moderate to strong correlations ranging from 0.60 for multiple species to 0.78 for SO₂ suggesting that one-third to one-half of the variance of analyzed pollutants is the same between the two sites such that there were common pollutant sources contributing at both sites.

4 DISCUSSION

4.1 Seasonal variations

Seasonal variations for key species are shown in Figure 3 that depicts the monthly average concentrations and the associated 75th and 99th confidence intervals calculated by bootstrapping the data (n=200). Seasonal variations for all of the analyzed pollutants are provided in Figure SI2. Nitrogen oxides, SO₂, NMHCs, UFPs in the first 3 size increments (20-70 nm) show typical seasonal cycles with maxima in the coldest seasons (Nov-Feb) and minima in the warmest months (Jun-Sept). Maxima in winter for such species are due to: (i) increase emissions for building space heating; (ii) the lower mixing layer heights and possibly strong thermal inversions, which limit the dispersion capacity of the atmosphere and trap the local emitted pollutants close to the ground (despite the higher wind speeds in winter), and (iii) the drop of ozone and hydroxyl radical and the consequent lower oxidation potential.

Average ozone concentrations show increased values in the warmest periods. OC and sulfate show higher concentrations in the warmest periods (Jul-Aug) along with SMPs in the accumulation mode (>100 nm). Increased concentrations of sulfate in warm seasons are the result of the increase in concentrations of atmospheric oxidants (mainly hydroxyl radical). The pattern is closely related to those of OX, air temperature and RH (Figure 3) as well as expected higher summer ammonia concentrations (Paulot et al., 2014). The sulfate/SO₂ ratios describes the increased speciation toward sulfate in summer: sulfate/SO₂ ratio range from values below 0.3 in winter to 0.8-1 from May to July.

The pattern of OC is strongly related to the formation of SOA, as evidenced by the comparison with the estimated OC_{sec} . Alternatively, OC_{pri} is more constant through the year, ranging from 0.7 to 1.1 μ gC m⁻³. In winter (December-February), OC_{sec} account for about 52% of total OC. In summer (June-August), OC_{sec} accounts for ~69-80% of total OC. This result is

comparable with Yu et al. (2004) for the Northeast U.S. (77 \pm 3%). These results are also consistent with secondary OC accounting for 40-50% of total OC during winter and 63-73% during summer at a NYC Bronx site (Rattigan et al., 2010). The pattern of OC_{sec} also well matches with the seasonal variations of SOA predicted by global atmospheric chemical transport models for U.S. (e.g., Liao et al., 2007). de Gouw et al. (2008) reported that organic matter in NE U.S. during summer is mostly the result of formation from anthropogenic VOCs precursors in urban plumes. The warmer seasons also experience increases in the emissions of biogenic VOCs from plants, mostly isoprene and terpenes, which are known strong SOA precursors (Carlton et al., 2009).

Particles in the 100-1000 nm range may originate from a variety of sources. Their seasonality is very similar to sulfate and, thus, may be also the result of secondary processes. This hypothesis is also partially supported by the data on the size resolved chemical composition of PM in northeastern U.S. For example, Cabada et al. (2004b) and Zhang et al. (2005) reported that sulfate is mostly distributed in the 100-1000 nm range at the Pittsburgh supersite.

Methane and CO show different behaviors. No methane data are available for May and June due to equipment problems. In the U.S., up-dated emission inventories (Miller et al., 2013; US EPA, 2015) report that enteric fermentation is the largest anthropogenic source of methane (\sim 26% of total emissions) followed by natural gas system and landfills. Methane showed higher concentrations in April probably the result of increased release from soil and wetlands as the ground warms. Carbon monoxide is primarily emitted by anthropogenic combustion (e.g., vehicular exhaust, domestic heating, industrial emissions, and biomass burning). It is also formed by oxidation of methane and NMHCs (Seinfeld and Pandis, 2006). The increases in winter can be primarily linked to space heating emissions and lower atmospheric dispersion similar to NO_x , SO_2 and NMHCs, while the peak in April-May may result from the oxidation of methane.

The remaining pollutants lacked significant seasonal cycles or their patterns were extremely variable. PM_{2.5} mass concentration showed slightly different patterns at QC and EP. PM_{2.5} mass concentration in QC exhibits a pattern comparable to sulfate, OC_{sec}, and particles >100 nm, consistent with significant secondary processes. However, PM_{2.5} lacked observable seasonality at EP. This discrepancy is due to the use of different methods. Although the adjustment method applied to TEOM concentrations at EP is able to approximate the data to

FRM, it should be noted that FRM does not capture the summer volatile PM. On the contrary, the FDMS system at QC collects more of the semi-volatile material, particularly in summer.

4.2 Weekly and daily cycles

Daily cycles (Figure 4 and Figure SI3) and weekly patterns (Figure SI4) are also the result of the interplay among emission strengths, photochemical processes, and micrometeorological factors. The daily cycles were computed after shifting the time to daylight saving time (DST) since sources like traffic and cooking will shift by an hour at the transition from standard to daylight savings and the reverse. Separating the data into two periods produces better-defined diurnal profiles. Generally, most of the analyzed species (NO, NO₂, NO_x, SO₂, CO, HMHCs, PM_{2.5}, sulfate, EC (both thermal and optical) and particles in the 20-200 nm ranges) show daily peaks corresponding to the rush hours of road traffic (6–9 am) and higher concentrations during weekdays. However, only some species (NO₂, SO₂, PM_{2.5}, and particles in the 30-200 nm range) also exhibit a second typical mode related to the evening rush hours (6-11 pm). For these latter species, morning and evening maxima are interspersed by lower concentrations that are assumed to be the result of: (i) lower emissions (less traffic); (ii) larger availability of atmospheric oxidants and (iii) higher mixing layer height and wind speeds and, consequently, enhanced atmospheric dispersion.

Ozone behaves oppositely since it depends on photochemical production during the day. Ozone exhibits higher concentrations on weekends as a result of lower emissions of NO typically because of lower heavy duty diesel traffic. The "weekend" effect is common in many airsheds (e.g., Pires et al., 2012; Huryn and Gough, 2014). The daily and weekly variations in ozone and anthropogenic emissions give rise to NO₂/NO_x ratios that show nitrogen oxide partitioning is shifted significantly toward NO during the morning rush hours. Patterns in the sulfate/SO₂ ratio evidence that SO₂ is always the prevailing species of sulfur. However, its transformation to sulfate is enhanced by higher concentrations of ozone and oxidants.

Total organic carbon in $PM_{2.5}$ does not show daily cycles, but it does have a strong weekly pattern with minima from Thursdays to Saturdays. Estimated primary and secondary OC show peculiar cycles: OC_{pri} has patterns similar to other primary pollutants, such as NO and CO, i.e. highest concentrations during weekdays and in the morning rush hours; OC_{sec} presents maxima during weekends and two peaks of high concentrations during the day. The first peak is

in the afternoon correlated to the maximum solar irradiation and, thus, maximum concentrations of atmospheric oxidants (OX). The second peak occurs in the evening and overnight. This could indicate transport of secondary material from other locations (Strader et al., 1999) or enhanced condensation of semi-volatile OC material due to lower ambient temperatures. It can also be hypothesized that the oxidation of VOCs of anthropogenic and natural origins based on nitrate radical oxidation can drive this behavior.

4.3 Intra-site correlations and cross-correlations

Intra-site relationships among different analyzed pollutants were investigated thought the correlation coefficients. Results are provided as Figures SI5 and SI6 for QC and EP, respectively. Correlation matrices were also clustered by using hierarchical cluster analysis to group pollutants showing similar correlations and help the interpretation of results.

At QC and EP, most variable pairs show moderate to strong correlations. At QC, the cluster analysis reveals that correlations may be split into 3 main groups. The first group links positive correlation among particles in the smaller ranges (20-100 nm) with pollutants directly emitted by combustion sources in urban environments (NO, SO₂, CO, NMHCs and EC (both thermal and optical EC)), methane and NO₂. The presence of cumulative variables (NO_x and THC) is not surprising since NO₂ accounts for most (68%) of NO_x and methane for most (92%) of THC. The second cluster shows positive correlations among coarser SMPs (>100 nm), PM_{2.5} mass concentration, and the main particulate species (OC and sulfate). Since sulfate is a main constituent of PM_{2.5} mass and OC_{sec} accounted for most of the OC mass, this cluster is likely the result of the secondary aerosol contributions to the PM_{2.5} mass concentrations. The presence of particles >100 nm along with OC and sulfate support their assignment to being secondary accumulation mode PM.

The last group consists only of ozone, and shows negative correlations with all other pollutants. Ozone plays a key role in reacting with primary pollutants and in the generation of secondary aerosol. However, ozone has weak correlations with variables in clusters 1 and 2. Thus, it can be hypothesized that ozone is driven primarily by local pollutants and chemistry, while sulfate and OC_{sec} PM have a non-local origin. This hypothesis will be investigated further using back-trajectory models.

The cross-correlation functions (CCF) were computed at ± 12 h lag to investigate the relationships among the pollutants and between pollutants and weather parameters in more detail. CCFs are calculated separately for the cold and warm seasons based on local daylight saving time (EST= standard time (Nov 1 2009 – Mar 14, 2010) and EDT= Eastern daylight time). The most important CCFs for QC and EP are presented in Figures SI7.

Generally, CCFs for pairs of variables shared by the two sites have similar profiles. CCFs for pairs of species mainly emitted by primary emission sources (NO, CO, SO₂, NMHCs, OC_{pri}, EC) are characterized by significant strong (>0.6) positive correlations near the 0 h lag, meaning that they have concurrent daily cycles, i.e. they are potentially emitted by the same set of sources and undergo to similar atmospheric processes. The correlations rapidly decrease by moving lags forward and backward, indicating that the relationships among primary pollutants is quite immediate.

The ozone-solar radiation relationship was used to investigate the photochemical production of ozone. It exhibited a peak at +2 and +3 h lags, i.e. the maximal ozone concentrations are delayed relative to the SR maxima. The oxidative chain driven by ozone were further analyzed in terms of the NO₂–O₃ and NO₂–NO pairs. Strong correlations (negative and positive, respectively) occur at lag 0 and reflect the fast kinetics of NO oxidation by ozone and hydroxyl radical (Finlayson Pitts and Pitts, 2000). The CCFs for the methane-ozone and sulfate-ozone pairs show weak correlations (0.3<r<0.6) over all lag values. Although the correlations are weak, the CCF for sulfate-ozone has positive values during the warm season and negative in cold periods reflecting the temperature dependence of the kinetics of S(IV) oxidation. It can be also hypothesized that negative CCF of sulfate-ozone in winter are due to the local sulfate production in NYC driven by the use of high sulfur-content home heating oils, whereas there is limited O₃ production.

CCFs for sulfate-PM_{2.5} and OC_{sec}-PM_{2.5} present moderate to strong (0.4-0.6) correlations over all lags. This result shows that sulfate and OC_{sec} drive the PM_{2.5} mass concentrations, and also indicates that secondary particle formation requires more than 12 h, i.e. such processes do not occur locally.

CCFs among SMPs and pollutants shared by the two sites are provided in Figures SI8 and SI9. Particles generally show significant positive correlations with NO, NO₂, SO₂ and PM_{2.5} at lag 0. However, correlations with NO, NO₂ and SO₂ are higher for particles in the 50-100 nm

range, while correlations with $PM_{2.5}$ increase by increasing the particle size. This latter result is largely expected, since large particles mostly contribute to the $PM_{2.5}$ mass.

4.4 Identification of local sources

CBPF plots are used to identify the potential local emission sources. The hourly air pollutant concentrations were matched with wind speed and direction values from the two weather stations. The criterion value was set at >75th percentile. All data were used, e.g., no wind calm hours were removed. The results are provided in Figure 5 for QC and Figure SI10 for EP. For data collected at QC, CBPF were calculated separately for the cold and warm seasons split based on the seasonal time change (Figures SI11 and SI12).

Generally, CBPFs at QC and EP show similar profiles for most of the analyzed species. Such profiles remain constant through the cold and warm seasons. However, higher probabilities of concentrations >75th percentile are found during the cold season for all of the locally emitted pollutants. In addition, there is less conversion of primary pollutants during winter due to reduced photochemical processes. High probabilities of NO, NO₂, CO, NMHCs, OC_{pri}, EC and SMP in the 30-100 nm range were attained during calm wind periods or under low/moderate wind speed (<5 m s⁻¹) with no prevailing direction. This pattern is compatible with the dilution effect driven by winds and the consequent advection of fresh air masses, and is also explained by the inverse relationships of the daily cycles between wind speed and primary pollutants (Figure 4). The highest wind speeds were observed in the afternoon and were concurrent with minimal concentrations of the local-emitted species. These results further support attribution of NO, NO₂, CO, NMHCs, OC_{pri}, EC and SMP in the 30-100 nm range to local emissions and/or fast atmospheric processes.

SO₂ shows increased probabilities with moderate westerly winds, i.e. toward the lower end of Manhattan. This direction is likely the result of emissions from central heating systems of the large buildings In NYC, the primary fuels used for space and water heating are fuel oils and natural gas. A large number of small residential and commercial buildings use distillate oil No. 2, while many larger buildings burn the less refined No. 6 (residual oil) with higher emission factors (US EPA, 2010; Kheirbek et al., 2014) or a mixture of Nos. 2 and 6 to produce No. 4 oil. In 2009-2010, No. 2 oil contained 0.2% sulfur (by weight), while the content in Nos. 4 and 6 was higher (0.3%). The annual building energy consumption by block area in NYC in 2009 reported

by Howard et al., 2012 shows that Manhattan has very high energy consumptions with respect to other boroughs and it is therefore likely to emit higher amounts of SO₂. However, shipping is another known strong source of SO₂ (e.g., Isakson et al., 2001; Eyring et al., 2005) and the CBPF is compatible with emissions from the container port at the Port of Elizabeth, NJ.

Methane, $PM_{2.5}$, sulfate, OC_{sec} and particles >100 nm exhibit high probabilities for moderate (4-7 m s⁻¹) winds blowing from SW. The highest probabilities are attained for air masses from Brooklyn. However, the relative high wind speeds associated with the CBPF indicate that the advection of air masses over Manhattan has the potential for raising their concentrations. Despite sulfate and OC_{sec} are related to secondary processes and, thus, may have a probable regional origin, it can be hypothesized that domestic heating and traffic may also contribute to their concentrations on a local scale. Ozone show increased probabilities for strong (>5 m s⁻¹) winds blowing from SSE. Thus, high ozone concentrations are observed when the winds blow from directions not heavily impacted by traffic, i.e. when NO is not sufficient to titrate out the O₃.

CBPF for the smallest particles (20-30 nm) show a different pattern. At QC, the higher concentrations are for winds blowing from NW, while the probability for calm wind periods is very low (<0.3). On the contrary, CBPF at EP shows a profile similar to other primary pollutants (moderate winds from W). The pattern at QC is not found for any other species and indicates a source of ultrafine particles in the nucleation mode toward NW. Potential emissions from high density of urban areas and the presence of main roads (e.g., Long Island Expressway, Van Wyck and Whitestone Expressways, Grand Central Parkway) are compatible with this result. High-density urban areas and frequently congested main roads are widespread in Queens. For example, the Long Island Expressway generally experiences the highest traffic volumes in Queens County, in excess of 200,000 vehicles day⁻¹ and it is frequently congested. As a consequence, minor roads in the area are also frequently congested. In addition, a large number of school buses moves in the area due to the College and other schools.

In addition to road traffic as a major source, the nucleation mode particles may be attributable to the emissions from LaGuardia Airport, approximately 5 km NW of QC. Aircraft emit ultrafine particles in the nucleation range (Masiol and Harrison, 2014; Lobo et al., 2015), thus airports are known sources of UFPs. For example, Hudda et al. (2014) reported that emissions from the Los Angeles International Airport increase PNC 4-fold at 10 km downwind,

while Keuken et al. (2015) reported that the particle number size distributions downwind of Schiphol Airport (The Netherlands) was dominated by ultrafine (10 to 20 nm) particles.

LaGuardia Airport is one of three commercial airports for NYC and had ~379,000 plane

movements in 2010.

4.5 Potential distant sources

The potential locations of distant sources were evaluated through PSCF analyses. The hourly air pollutant concentrations were matched with the back trajectories. The criterion value was set to concentrations >75th percentile. The PSCF surfaces were smoothed to decrease the noise due to the uncertainties of trajectories and to enhance the interpretability of maps. Results are shown in Figure 6 for QC and Figure SI13 for EP. For QC, the PSCF values were calculated separately for the cold and warm seasons (Figures SI14 and SI15).

Results show that continental eastern U.S. particularly the upper Ohio River Valley region was the highest probability source area for almost all the pollutants having likely regional origins, including SO₂, CH₄, PM_{2.5}, sulfate, OC_{sec}, and particles >50 nm. These patterns are almost constant during the cold and warm seasons. The probabilities for NO₂, SO₂, CO, NMHCs, OC_{pri}, EC and particles in the 50-100 nm range are generally less than 0.5. The maps do show probabilities >0.5 for methane, PM_{2.5}, sulfate and particles >100 nm. The maps for both sites clearly show that the potential effects of regional transports rise with increasing particle dimensions, with the highest probabilities for particles in the >200 nm range.

PSCF maps for particles >50 nm corresponded with the modeled emissions from the major UFP sources in the Eastern U.S. during summer as reported by Posner and Pandis (2015). The PSCF maps for particles in the 20-50 range do not show any pattern. The lack of any significant PSCF probability for finest particles means that they originate primarily from local sources and are not transported over long distances.

5. CONCLUSIONS

This study has provided direct information on the concentrations and trends of key air pollutants over high densely urbanized areas of NYC and Long Island. Primary and secondary OC are also estimated. Results enable some general considerations that can be summarized as follows:

• The seasonal, weekly and diurnal cycles of pollutants primarily emitted by anthropogenic sources (NO, SO₂, CO, NMHCs, OC_{pri}, EC and 30-100 nm particles) are strongly shaped by emission factors, human habits, photochemical processes and weather. Typical sinusoidal seasonality with maxima during cold seasons and uni/bimodal daily cycles related to rush hours are found for most pollutants. Nitrogen dioxide also follows this behavior due to the fast oxidation of NO. Intra-species analysis by means of cross-correlations has also pointed out the strong relationships among primary air pollutants by showing high positive correlations at 0 h lag. Primary pollutants increase during wind calm periods or under slow wind regimes and the CBPF analysis has evidenced that they are emitted locally. In addition, no significant effects of regional transports of primary species are evidenced from PSCF analysis. Results indicate different potential local sources: road traffic is the most probable source of nitrogen oxides, CO, NMHCs, OC_{pri}, EC and 30-100 nm particles, while domestic heating burning fuel oils and shipping for SO₂.

- PM_{2.5} and coarser particles are strongly related to secondary processes, as evidenced by the comparison with sulfate and estimated OC_{sec}. They show comparable seasonal and diurnal profiles, high correlations and relatively high lagged correlations over long periods (+/- 12 h). Also, CBPF results show that the advection of regional air masses has a potential effect in raising their concentrations, while PSCF pointed out the Eastern continental U.S. as most probable remote source area.
- Both local and regional sources can be found for UFPs in the nucleation range (<30 nm).
 Results indicate that LaGuardia airport is a potential local source at Queens, while PSCF analysis does not reveal any dominant remote source area.
- Ozone is linked to the photochemical production and is delayed (+2 h) with respect the maxima solar radiation. Correlations and diurnal cycles point out on its potential high effect as oxidant for primary species. Despite the large set of ozone-precursor emitted by anthropogenic sources within NYC boundaries, CBPF analysis has revealed that most O₃ is advected during high wind regimes, but not dominant source areas are evidenced by PSCF.

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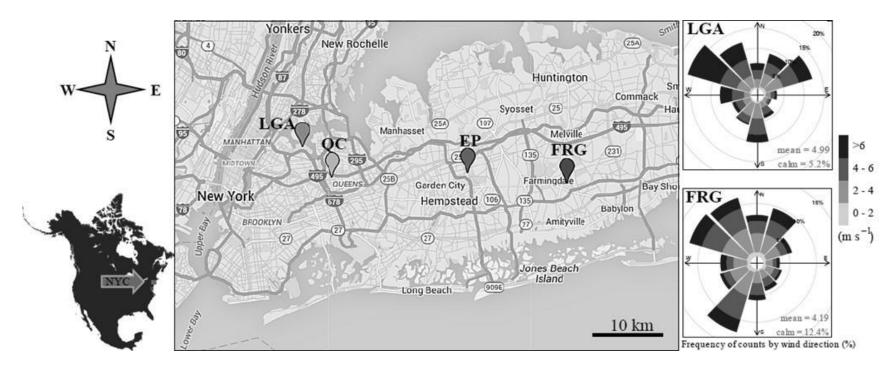


Figure 1. Map of the study area showing the two sampling sites (QC= Queens College; EP= Eisenhower Park) and wind roses for the sampling period (July 2009-July 2010) at the two weather stations (LGA= LaGuardia Airport; FRG= Republic Airport).

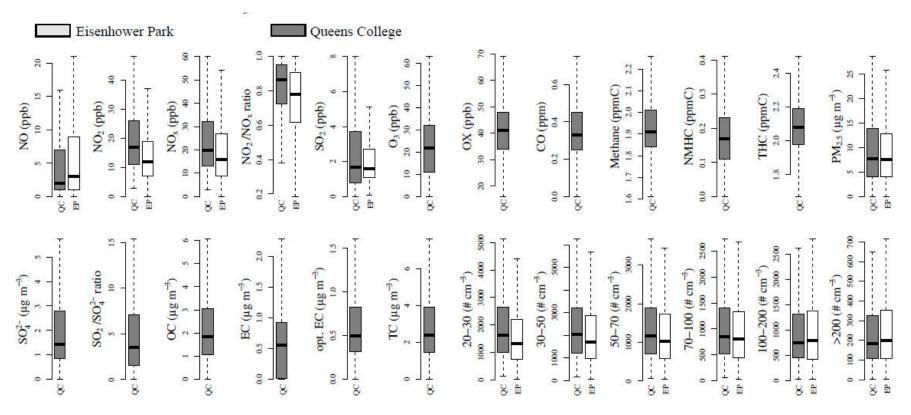


Figure 2. Boxplots of the analyzed pollutants. Lines= medians, boxes= th-75th percentile ranges, whiskers= ± 1.5 *inter-quartile ranges.

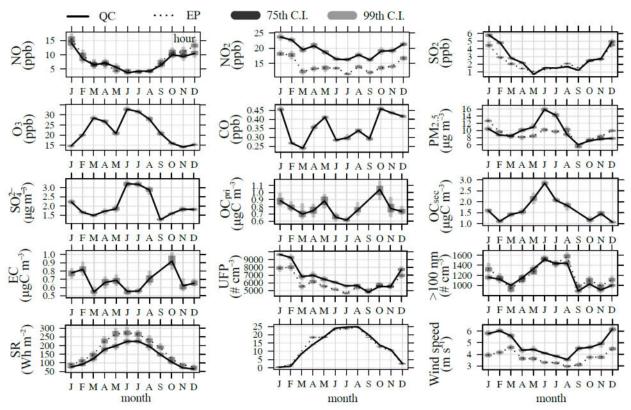


Figure 3. Seasonal variations of all monitored pollutants, derived parameters and some weather parameters. Each plot reports the monthly average concentrations as a filled line and the associated 75th and 99th confidence intervals calculated by bootstrapping the data (n=200).

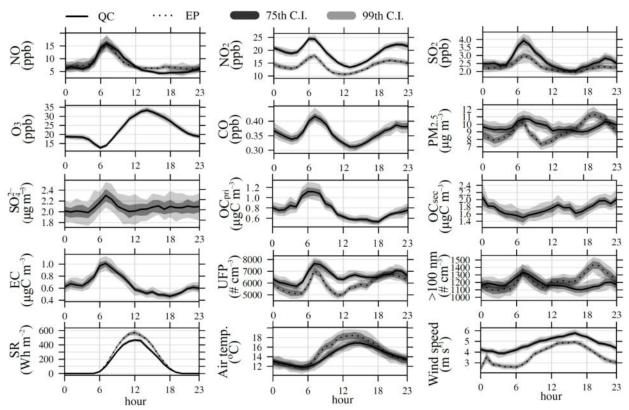


Figure 4. Diurnal variations in concentrations of measured pollutants. Each plot reports the average concentration as a filled line and the associated 75th and 99th confidence intervals calculated by bootstrapping the data (n=200).

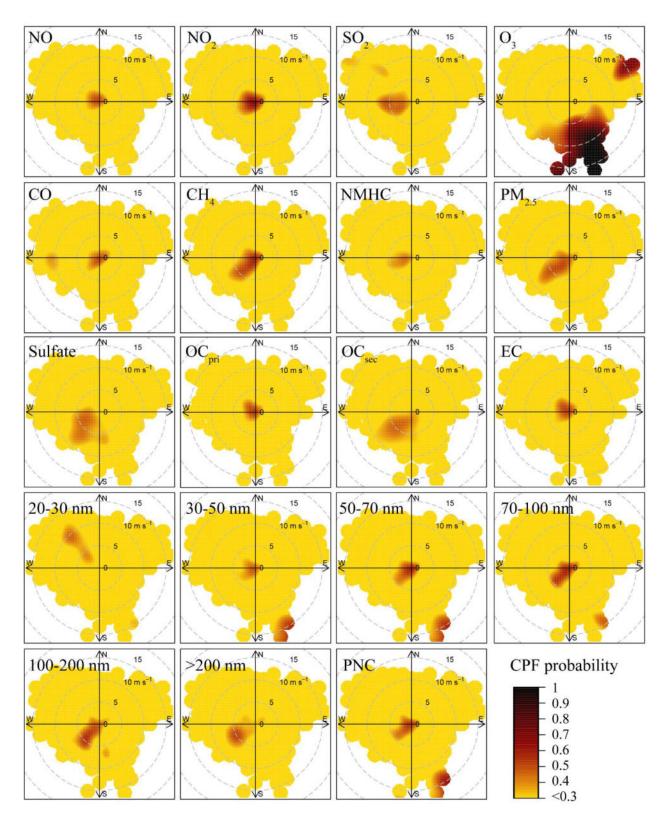


Figure 5. CBPF for pollutants measured at QC. CPF criteria: data from the whole sampling campaign; selected threshold: concentrations> 75th percentile.

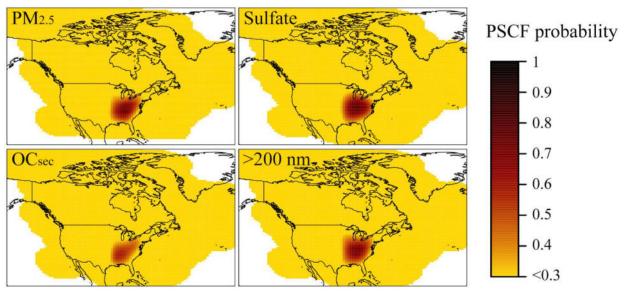


Figure 6. Smoothed PSCF maps for pollutants measured at QC during the whole sampling campaign. Selected threshold for PSCF: concentrations> 75th percentile.