| 1              | Levoglucosan and phenols in Antarctic marine, coastal and plateau aerosols   |
|----------------|--|
| 2              |  |
| 3              | Roberta Zangrando <sup>a*</sup> , Elena Barbaro <sup>a,b</sup> , Marco Vecchiato <sup>b</sup> , Natalie M. Kehrwald <sup>b</sup> , Carlo Barbante <sup>a</sup> , |
| 4              | Andrea Gambaro <sup>a,b</sup>  |
| 5              |  |
| 6              | <sup>a</sup> Institute for the Dynamics of Environmental Processes CNR, Dorsoduro 2137, 30123, Venice, Italy.  |
| 7              | <sup>b</sup> Department of Environmental Sciences, Informatics and Statistics, University of Venice, Ca'   |
| 8              | Foscari, Calle Larga Santa Marta 2137, 30123, Venice, Italy  |
| 9              |  |
| 10             |  |
| 11<br>12<br>13 | * Corresponding author phone: +39 041 2348945; fax: +390412348628; <u>mail:</u> roberta.zangrando@idpa.cnr.it.   |
| 14             |  |
| 15             |  |
| 16             |  |
| 17             |  |
| 18             |  |
| 19             |  |
| 20             |  |
| 21             |  |
| 22             |  |

#### 24 Abstract

25 Due to its isolated location, Antarctica is a natural laboratory for studying atmospheric aerosols and 26 pollution in remote areas. Here, we determined levoglucosan and phenolic compounds (PCs) at 27 diverse Antarctic sites: on the plateau, a coastal station and during an oceanographic cruise. Levoglucosan and PCs reached the Antarctic plateau where they were observed in accumulation mode 28 aerosols (with median levoglucosan concentrations of 6.4 pg m<sup>-3</sup> and 4.1 pg m<sup>-3</sup>, and median PC 29 concentrations of 15.0 pg m<sup>-3</sup> and 7.3 pg m<sup>-3</sup>). Aged aerosols arrived at the coastal site through 30 31 katabatic circulation with the majority of the levoglucosan mass distributed on larger particulates (24.8 pg m<sup>-3</sup>), while PCs were present in fine particles (34.0 pg m<sup>-3</sup>). The low levoglucosan/PC ratios 32 in Antarctic aerosols suggest that biomass burning aerosols only had regional, rather than local, 33 34 sources. General acid/aldehyde ratios were lower at the coastal site than on the plateau. Levoglucosan and PCs determined during the oceanographic cruise were 37.6 pg m<sup>-3</sup> and 58.5 pg m<sup>-3</sup> respectively. 35 Unlike levoglucosan, which can only be produced by biomass burning, PCs have both biomass 36 37 burning and other sources. Our comparisons of these two types of compounds across a range of 38 Antarctic marine, coastal, and plateau sites demonstrate that local marine sources dominate Antarctic 39 PC concentrations.

40

## 41 Keywords

- 42 Biomass burning
- 43 Levoglucosan
- 44 Phenolic compounds
- 45 Antarctica
- 46 Aerosols
- 47

### 49 Introduction

50 Biomass burning encompasses the combustion of living and dead vegetation, including wildfires, 51 prescribed burning (deforestation, shifting cultivation, agriculture waste) and domestic bio-fuel 52 combustion (such as in fireplaces, stoves etc) (Cheng at al., 2013). Humans intentionally and 53 accidentally ignite fires although volcanic activity and lightning also leads to forest fires. (Taylor 54 2010). Biomass combustion is the largest source of primary fine carbonaceous particles and the 55 second principal source of trace gases in the global atmosphere (Akagi at al., 2011).

56 Biomass burning aerosols influence the climate system by affecting the Earth's solar balance(IPCC 57 2013; Hobbs at al., 1997), acting as cloud condensation nuclei (Novakov and Corrigan, 1996; Vestin 58 at al., 2007) and influencing snow albedo (IPCC 2013; Flanner at al., 2007; Ramanathan and 59 Carmichael, 2008). However, the transport, evolution and sinks of many biomass burning aerosols 60 are not well understood. Here, we examine two classes of biomass burning tracers (levoglucosan and 61 phenolic compounds) in Antarctic plateau, coastal, and oceanic sites to determine how distance from 62 biomass burning source regions and subsequent transport and aging affects their concentrations and 63 size distribution.

64

65 Antarctica is surrounded by ocean, contains little to no biomass burning sources, lacks stable human 66 settlements, and therefore presents a natural laboratory for investigating biomass burning aerosols 67 after long range transport. We examine the specific biomarker levoglucosan (1,6 anhydro-β-D 68 glucopyranose) as it is an unambiguous product of cellulose combustion produced at temperatures of 69 approximately 250°C (Kuo at al., 2011). Here, we use levoglucosan as a reference biomass burning 70 tracer due to its specificity and high emission factors (Iinuma at al., 2007; Oros at al., 2006; Oros and 71 Simoneit 2001a; Oros and Simoneit 2001b). Although levoglucosan can degrade in the atmosphere 72 by reacting with OH (Hennigan at al., 2010; Hoffmann at al., 2010; Kessler at al., 2010), NO<sub>3</sub> and 73 SO<sub>4</sub> (Hoffmann at al., 2010), the high concentrations injected into smoke plumes suggests that 74 enough remains to allow using levoglucosan as a biomass burning tracer (Hoffmann at al., 2010). In 75 Arctic aerosols levoglucosan was determined both in conditions influenced by (Stohl at al., 2006; 76 Stohl at al., 2007) and not influenced (Fu at al., 2009; von Schneidemesser at al., 2009; Yttri at al., 77 2014; Zangrando at al., 2013) by wildfires, while in Antarctica studies only observe levoglucosan in 78 marine aerosols (Hu at al., 2013). Ice core (Gambaro at al., 2008; Kawamura at al., 2012; Legrand at 79 al., 2007; Yao at al., 2013) and snow pit (Hegg at al., 2010; Kehrwald at al., 2012) studies demonstrate 80 that levoglucosan can reconstruct past biomass burning over annual to millennial timescales (Zennaro 81 at al., 2014) in polar locations.

82 While levoglucosan records cellulose burning, this marker alone cannot determine what type 83 of vegetation burned to produce the smoke aerosols. PCs in atmospheric aerosols may indicate the 84 types of burned plants. Methoxy phenols derive from lignin combustion. Lignin is a biopolymer 85 comprised of three different aromatic alcohols; p-coumaryl, coniferyl and sinapyl alcohols where 86 their proportions differ between the major plant classes. The degradation products from oxidation or burning of lignin are classified as coumaryl, vanillyl and syringyl moieties (Simoneit 2002). 87 88 Hardwood (angiosperm) lignin (Oros and Simoneit 2001b) is enriched in sinapyl alcohol precursors 89 so burning these plants principally produces syringyl and vanillyl moieties. In deciduous tree smoke 90 the main PCs produced include homovanillyl alcohol, vanillic acid, vanillin, and syringic acid. 91 Softwoods (gymnosperms) (Oros and Simoneit 2001a) contain high proportions of coniferyl alcohol 92 with minor components from sinapyl alcohol and burning produces primarily vanillyl moieties. The 93 dominant phenolic biomarkers in conifer smoke include vanillin, homovanillic acid, vanillic acid, and 94 homovanillyl alcohol. In grasses (gramineae) (Oros at al., 2006) p-coumaryl alcohol is the dominant 95 lignin unit not prevalent in softwood and hardwood. Other significant products from burning grasses 96 are acetosyringone, syringic acid, vanillin and vanillic acid. Methoxy phenols degrade in the 97 atmosphere, where 2-methoxyphenol (guaiacol) and its isomers in the gas-phase react with OH

hydroxyradicals (Coeur-Tourneur at al., 2010), while phenols react with  ${}^{3}C^{*}$  (aromatic carbonyl) (Smith at al., 2014) and some methoxy phenols in particulate matter react with O<sub>3</sub> (Net at al., 2011), NO<sub>3</sub> (Liu at al., 2012),  ${}^{3}C^{*}$ (Yu at al., 2014), OH (Li at al., 2014; Yu at al., 2014), and UV (Li at al., 2014).

102 Most previous determinations of PCs in aerosols were performed in zones close to residential 103 areas using biomass burning in domestic heating (Bari at al., 2010; Bari at al., 2011; Dutton at al., 104 2010; Dutton at al., 2009; He at al., 2010; Simpson at al., 2005; Ward at al., 2011) or else in zones 105 heavily impacted from wildfire smoke (Ward at al., 2006). PCs occur in high concentrations near these biomass burning sources, ranging from 10s to greater than 10,000 pg m<sup>-3</sup> (Bari at al., 2010; Bari 106 107 at al., 2011; Dutton at al., 2010; Dutton at al., 2009; He at al., 2010; Simpson at al., 2005; Ward at 108 al., 2011). In the Arctic, PCs have considerably lower concentrations with mean values (for particle sizes of 10  $\mu$ m to <0.49  $\mu$ m) of 14 pg m<sup>-3</sup> (Zangrando at al., 2013). Several studies determine PCs in 109 110 ice and snow collected in Arctic areas (Hegg at al., 2010; Kawamura at al., 2012; McConnell at al., 111 2007), suggesting their applicability to Antarctic sites.

112 This work determines levoglucosan and PCs including vanillic acid (VA), isovanillic acid 113 (IVA), homovanillic acid (HA), syringic acid (SyA), vanillin (VAN), syringaldehyde (SyAH), ferulic 114 acid (FA), p-coumaric acid (PA) and coniferyl aldehyde (CAH) in three different Antarctic 115 environments in order to investigate how transport affects the concentrations, evolution and sinks of 116 these compounds in aerosols. We examine the concentrations and particle size distributions of 117 biomass burning tracers in remote aerosols at the Concordia Station (Dome C) on the East Antarctic 118 plateau during 2011-2012, 2012-2013, the coastal Mario Zucchelli Station in 2010-2011, and marine aerosol samples collected during the R/V Italica oceanographic cruise in the Southern Ocean in 2012 119 120 (Fig. 1).



- 122
- 123 Fig 1. Map reporting the Antarctic sites studied.
- 124

# 125 Experimental Section

#### 126 **Reagents and standard solutions**

127 HPLC/MS-grade methanol (MeOH) and acetonitrile (ACN) were purchased from Romil LTD (Cambridge, U.K.). The ultrapure water (18.2 MQ cm, 0.01 TOC) was produced by a Purelab Flex 128 (Elga, High Wycombe, U.K.) and formic acid (98%) was obtained by Fluka (Sigma Aldrich, Buchs, 129 Switzerland). Levoglucosan (purity 99%), vanillin (VAN) ( $\geq$  98%), syringic acid (SyA) ( $\geq$  95%), 130 131 homovanillic acid (HA) ( $\geq$  98%), isovanillic acid (IVA) (97%), *p*-coumaric acid (PA) ( $\geq$  98%), coniferyl aldehyde (CAH) (98%), were purchased from Sigma Aldrich, vanillic acid (VA) ( $\geq$  97%), 132 syringaldehyde (SyAH) ( $\geq$  97%), ferulic acid (FA) ( $\geq$ 99%) from Fluka. Levoglucosan<sup>13</sup>C<sub>6</sub>(98%) 133 isotopic enriched,  $\geq$  98% chemical purity) from Cambridge Isotope Laboratories Inc. (Andover, MA), 134 vanillic acid <sup>13</sup>C<sub>1</sub>(98% isotopic enriched,  $\geq$  98% chemical purity) and vanillin <sup>13</sup>C<sub>6</sub>(98% isotopic 135 136 enriched,  $\geq$  98% chemical purity) from Sigma Aldrich.

137

#### 138 Aerosol sampling

Aerosols were collected using a TE-6070, PM10 high volume air sampler (average flow 1.21 m<sup>3</sup> min<sup>-</sup> 140 141 <sup>1</sup>) with a Model TE-235 five stage high volume cascade impactor (Tisch Environmental Inc., Cleves, 142 OH) equipped with a high volume back-up filter (Quartz Fiber Filter Media 8" x 10" and 5.625" x 143 5.375" Slotted Quartz Fiber) for collecting particles in the following size ranges:  $10.0 - 7.2 \,\mu\text{m}$ , 7.2  $-3.0 \,\mu\text{m}, 3.0 - 1.5 \,\mu\text{m}, 1.5 - 0.95 \,\mu\text{m}, 0.95 - 0.49 \,\mu\text{m}, < 0.49 \,\mu\text{m}$ . Collection time was 10 days 144 resulting in a total air volume of ~15,000 m<sup>3</sup> per sample. Using this aerosol sampler, five size 145 146 segregated aerosol samples were collected at the Faraglione Camp ( $74^{\circ} 42' \text{ S} - 164^{\circ} 06' \text{ E}, 57 \text{ m asl}$ ), approximately 3 km south of the Mario Zucchelli Station in Victoria Land, from November 29, 2010 147 148 to January 18, 2011. In the East Antarctic plateau (75° 06' S – 123° 20' E) approximately 1 km south-149 west of the Dome C building, four aerosol samples were acquired from December 19, 2011 to January 150 28, 2012 while five airborne samples were collected from December 7, 2012 to January 26, 2013. 151 A TE 5000 High Volume Air Sampler (Tisch Environmental Inc., OH) was used to collect thirteen 152 TSP (Total Suspended Particles) samples on a circular quartz fiber filter (SKC Inc., Eighty Four, To-153 13 model).Seven samples were collected over the Ross Sea (Antarctica) on the R/V Italica from 154 January 13 to February 19, 2012 (Table S1). In order to avoid contamination from the ship's exhaust 155 during the oceanographic cruise, the air samples were automatically controlled by a wind sector to

relative wind was more than 1 m s<sup>-1</sup>. Collection time was approximately five days and varied due to wind direction and the cruise events resulting in air sampling volumes between 511 and 2156 m<sup>3</sup>.

start sampling only when the relative wind direction ranged from -135° to 135° of the bow and the

All filters were pre-combusted (4 h at 400°C in a muffle furnace) and wrapped in two aluminum foils before sampling and stored in aluminum at -20°C after sampling until analysis. Blank samples were collected by loading, carrying and installing the filter holder in the instrument with a closed air pump.

162

156

#### 163 Sample processing, quality control and instrumental analysis

Our aim was to determine all phenolic compounds and levoglucosan in the Antarctic samples using an extraction in water proposed by Zangrando et al. (Zangrando at al., 2013). However, we modified the preanalytical protocol by using ice during the extraction in the ultrasonic bath in order to reduce the volatility and degradation of phenolic compounds. This modification allowed the extracting levoglucosan and all phenolic compounds plus VAN in water. In the previous version of method (Zangrando at al., 2013) SyAH and PA were extracted in methanol and VAN was not present.

170 In order to avoid contamination from laboratory air particles, samples were handled under a class 100 171 laminar flow bench. Each quartz fiber filter was cut in half using stainless steel scissors that were previously washed with methanol. Filter pieces were placed into 50 mL conical flacks and spiked 172 173 with internal standard solutions. Slotted quartz fiber supports and circular quartz fiber filters were spiked with 140  $\mu$ L of isotopically-labelled<sup>13</sup>C<sub>6</sub> levoglucosan (4  $\mu$ gmL<sup>-1</sup>), 70 of <sup>13</sup>C<sub>6</sub>vanillin (1 $\mu$ gmL<sup>-1</sup>) 174 <sup>1</sup>) and <sup>13</sup>C<sub>1</sub>vanillic acid (1µgmL<sup>-1</sup>) standard solutions and extracted with 5 mL and then 2 mL of 175 176 ultrapure water by ultrasonication. Each piece of background filter was spiked with 500 µL of labeled levoglucosan and 300 µL of labeled VA and VAN. Vanillin was determined using <sup>13</sup>C<sub>6</sub>vanillin as 177 178 internal standard. The filters were then extracted with 25 mL and then 5 mL of ultrapure water. The 179 extracts were combined and filtered through a 0.45 µm PTFE filter in order to remove particulates 180 before instrumental analysis.

181 Each type of sampling filter (round, backup and slotted) and each analyte were validated. The 182 trueness, reproducibility and efficiency of the sample preparation procedure (yield %) resulted in a percent error of 10%, CV% less than 10% and a yield% greater than 50% for each compound. This 183 184 unique procedure permits determining levoglucosan and PCs. For PCs determination the instrumental 185 method in Zangrando et al. (Zangrando at al., 2013) was upgraded to add monitored transitions for 186 VAN: m/z 150.9/135.8 (used for quantification) and m/z 150.9/92.0. Good linearity was observed for vanillin between 0.01 and 100 pg  $\mu$ L<sup>-1</sup>, R<sup>2</sup> =0.99. For both Declustering Potential and Entrance 187 Potential Declustering potential were -50 V and -10 V. Collision Energy and Cell Exit Potential were 188 -18 V and -11 V for *m/z* 150.9/135.8 and -27.5 V and -6 V for *m/z* 150.9/92.0 respectively. 189

Instrumental analysis of levoglucosan compounds was performed by HPLC/(-)ESI-MS/MS as
described in Perrone et al. (Perrone at al., 2012).

192

#### 193 Back-trajectory calculation

Back-trajectories for Mario Zucchelli Station, Dome C and R/V Italica were computed using the Hybrid Single Particle Lagrangian Integrated Trajectory (HYSPLIT) transport and dispersion model(Draxler and Rolph, 2013). All back-trajectories use NCEP/NCAR Global Reanalysis Data. The vertical velocity model used for Mario Zucchelli Station data incorporated vertical motion while we employed the isoentropic model for Dome C air mass analyses, as suggested by Stohl et al (Stohl and Sodemann, 2010).

200

201 In order to highlight the main pattern of air masses, 240-hour normal back-trajectories beginning 500 202 m above ground level (AGL) at Mario Zucchelli Station and Dome C were calculated during each 203 sampling campaign period. Four runs were computed for every sampling day starting every six hours 204 and the resulting trajectories were mean-clustered into 6 groups. We evaluated how altitude 205 influences the HYSPLIT back trajectories by calculating the cluster mean back trajectories at Mario 206 Zucchelli Station and Dome C for four different heights: 10 m, 100 m, 500 m and 1000 m. Regardless 207 of the atmospheric height, the resulting air mass clusters have similar geographical trajectories (Fig. 208 S1A-1C) but where the percentages of air masses in each cluster change. Because the aim of this 209 work is to study long range transport we used an atmospheric height of 500 m (NOAA) The mean 210 mixed layer height is 200-400 m agl at Dome C (Argentini at al., 2005) and at the coastal Halley site 211 the boundary layer height is usually up to 200 m (Saiz-Lopez at al., 2008). For the oceanographic 212 cruise, we performed trajectory matrices in order to simulate the ship track. We computed 5-day back-213 trajectories for each oceanographic 24-h sampling event.

#### 215 **Results and discussion**

Levoglucosan and Phenolic Compounds in size segregated particles over the Antarctic plateau 217 218 In order to investigate background levoglucosan levels over the central Antarctic plateau, we 219 determined levoglucosan concentrations during sampling campaigns in 2011-2012 and 2012-2013 220 resulting in respective median values of 6.4 pg m<sup>-3</sup> and 4.1 pg m<sup>-3</sup> (Fig 2A, 2B). Dome C levoglucosan 221 concentrations were primarily present in accumulation mode aerosols (< 1  $\mu$ m), accounting for 76% 222 (2011-2012) and 77% (2012-2013) of the total levoglucosan concentrations determined in PM<sub>10</sub>. The Dome C levoglucosan concentrations are significantly lower than mean Summit, Greenland (lat 72° 223 36' N lon 38° 25' W, 3231 m asl) atmospheric levoglucosan concentrations of 320 pg m<sup>-3</sup> (ranging 224 from 170 to 830 pg m<sup>-3</sup>)(von Schneidemesser at al., 2009). This order of magnitude difference in 225 226 central Antarctic versus central Greenland levoglucosan concentrations may be due to the greater 227 distance of Dome C from any biomass combustion source.





Fig 2: Size distributions of levoglucosan concentrations in samples collected at the Concordia Station
during the austral summers 2011-12 (A) and 2012-13 (B) at the Mario Zucchelli Station (Antarctica)
during 2010-2011 (C). Particle size dimesions (S1: 10-7.2 μm; S2: 7.2-3.0 μm; S3: 3.0-1.5 μm; S4:
1.5- 0.95 μm; S5: 0.95-0.49 μm; Backup Filter (BF): <0.49 μm).</li>

234

235 To the best of our knowledge, this study represents the first time PCs were determined in Antarctic plateau aerosols. We determined nine PCs: VAN, VA, HA, PA, SyAH, SyAH, IVA, CAH, FA. 236 237 Isovanillic acid concentrations were below detection limits in all samples in each campaign. The 238 Antarctic plateau samples from Dome C in 2011-2012 and 2012-2013 resulted in median total PC atmospheric concentrations of 15.0 pg m<sup>-3</sup> and 7.3 pg m<sup>-3</sup> respectively (Fig 3A, 3B). The main 239 240 compounds present in all sample size fractions were VAN, VA and PA. In 2011-2012 these 241 compounds respectively represented 47%, 26% and 14% of the total PCs in the  $< 0.49 \ \mu m$  size fraction while in 2012-2013 these compounds accounted for 58%, 12% and 24%. HA was absent in 242 243 all Dome C samples.



Fig 3: Size distributions of phenolic compound concentrations in samples collected during the austral summers 2011-12 (A) and 2012-13 (B) at the Dome C and 2010-2011 (C) at the Mario Zucchelli Station (Antarctica). Particle size dimension s(S1: 10-7.2  $\mu$ m; S2: 7.2-3.0  $\mu$ m; S3: 3.0-1.5  $\mu$ m; S4: 1.5- 0.95  $\mu$ m; S5: 0.95-0.49  $\mu$ m; Backup Filter (BF): <0.49  $\mu$ m).

266

|   | The majority of FCs in Dome C occur in the fine fraction, where this distribution is consistent   |
|---|---|
| 252   | with long range transport (Roiger at al., 2012; Fattori at al., 2005; Jarvinen at al., 2013; Udisti at al.,   |
| 253   | 2012) (Figs 3A-B). PCs in fine particles (< 1 $\mu$ m) accounted for the 47% of total concentration in the  |
| 254   | 2011-2012 samples and 75% of the 2012-2013 samples. Even though the majority of PCs in the Dome   |
| 255   | C aerosols occur in the fine fraction, PCs are also present in the coarse fraction in 2011-2012. This   |
| 256   | behavior is consistent with their physico-chemical properties: PCs are semivolatile species   |
| 257   | (Hawthorne at al., 1992; He at al., 2010) that can volatilize and recondense on coarse particles  |
| 258   | (Herckes at al., 2006; Wang at al., 2011) during long-range transport.  |
| 250   |   |
| 239   | Cluster mean back trajectories help determine why PCs may be more present in the coarse   |
| 259<br>260  | Cluster mean back trajectories help determine why PCs may be more present in the coarse fraction in 2011-2012 (Fig. S2 and S3). The majority ,71% (2011-2012) and 68% (2012-2013), of the   |
| 259<br>260<br>261   | Cluster mean back trajectories help determine why PCs may be more present in the coarse fraction in 2011-2012 (Fig. S2 and S3). The majority ,71% (2011-2012) and 68% (2012-2013), of the air masses reaching Dome C only travelled over plateau locations during the sampling periods. The   |
| <ul><li>259</li><li>260</li><li>261</li><li>262</li></ul>   | Cluster mean back trajectories help determine why PCs may be more present in the coarse fraction in 2011-2012 (Fig. S2 and S3). The majority ,71% (2011-2012) and 68% (2012-2013), of the air masses reaching Dome C only travelled over plateau locations during the sampling periods. The remaining air masses travelled over the Southern Ocean (29% in 2011-2012 and 32% in 2012-2013).   |
| <ul> <li>259</li> <li>260</li> <li>261</li> <li>262</li> <li>263</li> </ul>                           | Cluster mean back trajectories help determine why PCs may be more present in the coarse fraction in 2011-2012 (Fig. S2 and S3). The majority ,71% (2011-2012) and 68% (2012-2013), of the air masses reaching Dome C only travelled over plateau locations during the sampling periods. The remaining air masses travelled over the Southern Ocean (29% in 2011-2012 and 32% in 2012-2013). The air masses originating over the Southern Ocean reached Dome C much more quickly in 2011-  |
| <ul> <li>259</li> <li>260</li> <li>261</li> <li>262</li> <li>263</li> <li>264</li> </ul>              | Cluster mean back trajectories help determine why PCs may be more present in the coarse fraction in 2011-2012 (Fig. S2 and S3). The majority ,71% (2011-2012) and 68% (2012-2013), of the air masses reaching Dome C only travelled over plateau locations during the sampling periods. The remaining air masses travelled over the Southern Ocean (29% in 2011-2012 and 32% in 2012-2013). The air masses originating over the Southern Ocean reached Dome C much more quickly in 2011-2012, requiring only approximately 36 hours to reach Dome C, versus the 2012-2013 air masses that   |
| <ul> <li>259</li> <li>260</li> <li>261</li> <li>262</li> <li>263</li> <li>264</li> <li>265</li> </ul> | Cluster mean back trajectories help determine why PCs may be more present in the coarse fraction in 2011-2012 (Fig. S2 and S3). The majority ,71% (2011-2012) and 68% (2012-2013), of the air masses reaching Dome C only travelled over plateau locations during the sampling periods. The remaining air masses travelled over the Southern Ocean (29% in 2011-2012 and 32% in 2012-2013). The air masses originating over the Southern Ocean reached Dome C much more quickly in 2011-2012, requiring only approximately 36 hours to reach Dome C, versus the 2012-2013 air masses that travelled between the Southern Ocean and Dome C in four to seven days. This difference in transport |

267 likely deposited PCs during transport as tropospheric particles between 5 -10 µm have atmospheric
268 lifetimes of only 1-2 days (Petzold and Karcher, 2012)

time may have influenced the particle size distribution. The slower-moving air masses in 2012-2013

Although the majority of the Dome C samples contain levoglucosan concentrations less than 10 pg m<sup>-3</sup>, two samples (December 19-29, 2011 and December 27, 2012 – January 6, 2013) have relatively high levoglucosan concentrations and different size distribution patterns (Fig. 2A, 2B). Both of these samples may be influenced by particular sources. The December 19 - 29, 2011 sample simultaneously contains high levoglucosan (25.6 pg m<sup>-3</sup>) and PCs (34.4 pg m<sup>-3</sup>) (fig 2A-3A) compared to levoglucosan concentrations in all other 2011-2012 of 1.3 and 4.0 pg m<sup>-3</sup> and PC concentrations of 4.8-15.3 pg m<sup>-3</sup>. The circulation patterns during this time period (Fig. S4) demonstrate that air masses 276 reaching Dome C originated over the ocean and the back trajectory paths are similar to others 277 occurring during times with low levoglucosan and PCs concentrations. However, unusual combustion sources may influence this sample. The time interval of December 19-29, 2011 encompasses holiday 278 279 festivities at the Dome C base, possibly accounting for the increased biomass burning concentrations. 280 The December 27, 2012 – January 6, 2013 sample also contains a plausible particular source. An 281 Antarctic traverse team arrived at Dome C December 31, 2012 (PNRA). The traverse used smoke 282 bombs that incorporate lactose to produce the smoke. Lactose is a disaccharide sugar composed from glucose and galactose, which can produce levoglucosan during combustion (here at 36.4 pg m<sup>-3</sup>), but 283 due to the lack of lignin in smoke bombs, these bombs do not produce phenolic compunds. The low 284 concentration of PCs (13.4 pg m<sup>-3</sup>) in this sample supports this conclusion of a particular source (Fig. 285 3B). 286

287

# Levoglucosan and Phenolic Compounds in size segregated particles on the Antarctic coast 289

290 We sampled at the Faraglione Camp near Mario Zucchelli Station during austral summer 2010-2011 291 using a high volume impactor. Particulate matter collected at the coastal site during 2010-2011 contains median atmospheric levoglucosan concentrations of 24.8 pg m<sup>-3</sup> (ranging between 12.9-292 175.0 pg m<sup>-3</sup>), with a particle size distribution that is relatively enriched in the coarse fraction (Fig 293 294 2C). Due to the distance from biomass burning sources, Antarctic levoglucosan concentrations are 295 substantially lower than concentrations determined in other polar coastal sites such as summer concentrations in Ny Alesund (Svalbard) at Gruvebadet with mean concentrations of 65 pg m<sup>-3</sup> 296 (ranging from 4 to 682 pg m<sup>-3</sup>) (Zangrando at al., 2013), summer and winter concentrations at the 297 Zeppelin observatory(Yttri at al., 2014) (1020 pg m<sup>-3</sup> and 130 pg m<sup>-3</sup>), and Alert (Canada)(Fu at al., 298 2009) which has mean concentrations of 147 (ranging between  $3-1076 \text{ pg m}^{-3}$ ). 299

300 In the five samples collected in 2010-2011, PCs were predominantly present in the fine 301 fraction (<0.49  $\mu$ m). The median concentration of PCs in the <0.49  $\mu$ m fraction was 22.3 pg m<sup>-3</sup> 302 (range 14-73 pg m<sup>-3</sup>) while the total median concentrations in the particle size  $<10 \mu m$  was 33.7 pg m<sup>-3</sup> (range 25-90 pg m<sup>-3</sup>) (Table S2 and Fig 3C). The only other concentrations of PCs in polar 303 304 aerosols available in the literature were obtained at the coastal Arctic site of NyAlesund (Zangrando 305 at al., 2013). In order to compare the data collected in coastal Arctic and Antarctic sites, we excluded 306 vanillin from the Antarctic data as it was not determined at NyAlesund. The PC concentrations in 307 <0.49 µm represent 36% of the total PCs determined in the Ny Alesund samples with a median concentration of 19.0 pg m<sup>-3</sup> (range 3.6-132.8 pg m<sup>-3</sup>) (Zangrando at al., 2013). The main PCs present 308 309 in the Ny Alesund samples were VA, HA, PA and SyAH which represented 28%, 27%, 39% and 310 3.5% of the total PC concentrations, respectively (Zangrando at al., 2013). The Faraglione Camp 311 samples contained 54% of the total PC concentrations in the  $<0.49 \mu m$  fraction with a median total concentration of 6.6 pg m<sup>-3</sup> (range 4.2-18.1 pg m<sup>-3</sup>). The composition of the PCs in the  $< 0.49 \mu m$ 312 313 fraction of the Antarctic aerosol samples contained VAN (74%), VA (10%), HA (7%), PA (6%), 314 while SyA, CAH, SyAH represented only 4% of the total. The coastal samples contain HA (7%) 315 which was not present in Dome C particulate matter.

316

317 At Mario Zucchelli Station in 2010-2011, 36% of the total levoglucosan concentrations were present in the coarse fraction. However, the Dome C plateau levoglucosan concentrations occurred mainly in 318 319 accumulation mode particles where levoglucosan on coarse particles represented only 24% of the 320 total in 2010-2011 and 23% of the total in 2012-2013. This enhancement of levoglucosan 321 concentrations in the coarse fraction at Mario Zucchelli Station may result from the hygroscopic growth of particles. More than 95% of air masses descend from the cold, dry Antarctic plateau (King 322 323 and Turner, 1997), to reach the relatively more temperate and humid Mario Zucchelli Station coastal 324 site (Fig S5).

325

The literature demonstrates that hygroscopic properties exist in mixed organic and inorganic aerosols
containing levoglucosan (Svenningsson at al., 2006). Recent studies establish that VA and SyA do

not hygroscopicaly grow (Mochida and Kawamura 2004), although the reaction of phenols with  ${}^{3}C^{*}$ and OH produce highly oxygenated species (Li at al., 2014; Smith at al., 2014; Yu at al., 2014) with higher hygroscopicity (Li at al., 2014). At Aboa (Antarctica) Asmi et al.(Asmi at al., 2010) reported the high hygroscopicity of aerosols from continental air due to the presence of chemically evolved species carried by particulate matter.

333

Our results are surprising as we mainly observe PCs in fine particles (Fig 3C) while the volatility of 334 335 PCs suggests that they should be distributed on coarse particles (Herckes at al., 2006) during longrange transport. In addition, degradated hygroscopic compounds (Asmi at al., 2010) should also 336 337 increase particle size. The presence of PCs primarily in the fine fraction suggests a local Antarctic 338 source. Faraglione Camp is located on a cliff above the Ross Sea, where the ocean supplies humidity 339 and marine aerosols. The ocean is a possible local source of PCs by ejecting fine particles (<1µm) 340 containing organic compounds into the atmosphere primarily during periods of high biological 341 activity but also during periods of low primary productivity (Ault at al., 2013; O'Dowd and De Leeuw 342 2007; Schmitt-Kopplin at al., 2012).

343

PC ratios further support the idea that the coastal samples derive from a local source. The VA/VAN and SyA/SyAH (Table 1) ratios indicate of the oxidation of PCs (Net at al., 2011) and consequently the degree of aerosol transformation. Student t-tests demonstrate that mean VA/VAN and SyA/SyAH ratios were significantly lower at the coastal site than at the Dome C plateau location. These results were consistent for both sampling seasons with VA/VAN (p=0.0095 2011-2012; p=0.014 2012-2013), and SyA/SyAH (p=0.00019 2011-2012; p=0.00012 2012-2013) demonstrating the presence of more oxidized aerosols at Dome C.

351

352 Comparing PCs with levoglucosan in Antarctic aerosols helps determine if the PCs in particulate 353 matter are affected by biomass burning. Dry wood mass is composed of 25-30% lignin, while 354 cellulose and hemicellulose account for the remaining 40-50% and 20-30%, respectively(Oros and 355 Simoneit 2001a). As levoglucosan derives from the pyrolysis of cellulose and PCs derive from the 356 combustion of lignin, the larger proportion of cellulose in wood results in biomass burning injecting 357 more levoglucosan than PCs into smoke plumes. In general levoglucosan/VA, levoglucosan/VAN, 358 levoglucosan/SyA and levoglucosan/SyAH ratios (calculated for atmospheric aerosols affected by 359 biomass burning) range from approximately 10 to 1000 demonstrating higher concentrations of 360 levoglucosan with respect to PCs (Table 1 and references therein). In Antarctic sites, Mario Zucchelli 361 Station and Dome C, these ratios were up to 100-1000 times (Table 1) less indicating higher 362 concentrations of PCs than biomass combustion aerosols

363

Mario Zucchelli Station samples contained relatively consistent concentrations and particle size 364 distributions of levoglucosan (Fig 2C) and PCs (Fig 3C), except for the November 29, 2010 sample 365 that contains the highest concentrations (levoglucosan 175.0 pg m<sup>-3</sup> and PCs 90.3 pg m<sup>-3</sup>) of the 366 367 season. In this sample the majority of the levoglucosan is present in the fine fraction, where this high 368 concentration and size fraction distribution is similar to fresh particulate matter produced from 369 biomass burning. In wood combustion aerosols levoglucosan is often associated with particles less 370 than 1.1 µm (Agarwal at al., 2010; Herckes at al., 2006; Kleeman at al., 2008; Schkolnik at al., 2005) 371 and aggregates range from 120 nm to 1 µm (Mavrocordatos at al., 2002). Similarly PCs were observed 372 in fresh smoke mainly in fine particles (Herckes at al., 2006; Iinuma at al., 2007). This combination 373 of factors suggests the influence of a particular biomass burning source such as the neighbouring 374 Italian base where an incinerator was used to burn waste such as paper, cardboard, and unvarnished 375 wood, especially during November and December 2010-2011, as personally witnessed by the author.

376

# 377 Levoglucosan and Phenolic Compounds in total suspended particles over the Southern Ocean 378 We determined levoglucosan and PCs in aerosol samples collected over the Southern Ocean during 379 the R/V Italica research cruise (Table S1) from January 13 to February 19, 2012 during the trip to and

380 from Mario Zucchelli Station. The January 13-18 and January 18-23 samples were collected during 381 the journey from Littleton Harbor, New Zealand to Mario Zucchelli Station. The January 25-29, January 31- February 6, and February 7-9 samples were collected over the Ross Sea near the Antarctic 382 383 continent. The February 12-18 and February 18-19 samples were collected during the return trip to 384 New Zealand. We used a TSP sampler to collect fine particles, and the coarse fraction resulting 385 particles with a marine origin(O'Dowd and De Leeuw, 2007; O'Dowd at al., 2004) and from hygroscopic growth(Petzold and Karcher, 2012; Raes at al., 2000). Median levoglucosan 386 387 concentrations were 37.6 pg m<sup>-3</sup>, ranging from BDL to 224.1 pg m<sup>-3</sup> (Fig S6 and Table S2) and the median phenolic compound concentrations for all samples was 58.5 pg m<sup>-3</sup> (Fig S7 and Table S2). 388

389

The January 13-18 sample has the highest levoglucosan concentrations (224.1 pg m<sup>-3</sup>) of the marine 390 samples, and also contains high total PC concentrations of 107.4 pg m<sup>-3</sup> Back trajectories demonstrate 391 392 that these high concentrations may be due to the influence of fires occurring in New Zealand (Fig 393 S8A and S9) during this time period, where the smoke plumes were transported to the sampling area. 394 Low concentrations in the February 18-19 sample, which is also located in New Zealand coastal areas, 395 registered levoglucosan BDL and low PC concentrations of 16.6 pg m<sup>-3</sup>. These low levels can be 396 explained by the short sampling time (only two days for this sample versus five days for the other 397 samples).

398

The January 18-23 and February 12-18 samples were collected mainly in the Antarctic Convergence area distant from both New Zealand and Antarctica, with a marked presence of sea ice. Levoglucosan has atmospheric concentrations of 22.4 and 37.6 pg m<sup>-3</sup> and PCs of 5.4 pg m<sup>-3</sup> and 9.7 pg m<sup>-3</sup> respectively.

| 404 | January 25-29, January 31- February 6, and February 7-9 samples were collected over the   |
|-----|---|
| 405 | Ross Sea (Fig S10), and have a mean levoglucosan concentration of 73.5 pg m <sup>-3</sup> . Recently Hu et al.                              |
| 406 | (Hu at al., 2013) reported levoglucosan concentrations recorded during Southern Ocean research  |
| 407 | cruises of 4.8 ng m <sup>-3</sup> (range 1.1-18 ng m <sup>-3</sup> ) near the East Antarctica coast and 3.4 ng m <sup>-3</sup> (range 0.18- |
| 408 | 11 ng m <sup>-3</sup> ) near the West Antarctica shore. These data together with black carbon studies from the                              |
| 409 | Antarctic stations Syowa (Hara at al., 2010), Troll (Fiebig at al., 2009), Ferraz (Pereira at al., 2006),                                   |
| 410 | Halley (Wolff and Cachier 1998) and Ross Island (Murphey and Hogan 1992) demonstrate that South   |
| 411 | American biomass burning is a possible source of aerosols reaching Antarctica (Hara at al., 2010).  |
| 412 | However, 5-day back trajectories demonstrate that katabatic winds transport material from the interior                                      |
| 413 | of West Antarctica during the time period comprised by samples January 25-29, January 31- February  |
| 414 | 6, and February 7-9 (Fig. S8C-E). The major influence of katabatic winds conveying fine   |
| 415 | particles(Asmi at al., 2010; Pant at al., 2011) originating from the interior of the continent (Fig S11A),                                  |
| 416 | appears during the February 7-9 sample when we record low levoglucosan concentrations in the Ross   |
| 417 | Sea.  |

In addition to katabatic circulation, during individual days of the sampling periods January 25-27 and February 1, 2, and 5 the wind blew from the Ross Sea to the Ross Ice Shelf (Figs S11B). In the January 25-29 and January 31- February 6 samples, the higher levoglucosan concentrations can be explained by the intrusion of air masses passing the Southern Ocean (versus originating from the Antarctic interior) where these air masses may contain biomass burning aerosols (Hu at al., 2013).

424

PC concentrations are relatively high for the time periods when the cruise traveled near the Ross Ice shelf (January 25-29, 2012 and, January 31-February 6 2012) and are especially high when the research vessel sailed through the Ross Sea polynya (February 7-9, 2012), (Fig S7). HA (67%) was the most abundant phenolic compound in the Ross Sea samples followed by PA (22%), VA (8%), and FA (1%) and SyA (3%). We did not determine the aldehydes VAN and SyAH as field blanks for these compounds were higher than the amount sampled, suggesting high levels of these aldehydes in marine aerosols. The TSP sampler demonstrates that HA and PA are prevalent in marine aerosols, even including the coarse fraction. These marine coarse fraction aerosols help interpret the coastal coarse fraction data where HA and PA increased to 29% and 26%, showing an influence of marine aerosols on coastal particles.

435

#### 436 Conclusion

437

Here, we determined that levoglucosan can be detected in remote areas, even in sites as distant from biomass burning sources as Dome C, East Antarctica. Our results indicate that the biomass burning tracer levoglucosan reached the inner Antarctic plateau through long-range transport and was present in accumulation-mode aerosols. At the coastal site levoglucosan was substantially present on coarse particles created by hygroscopic growth. During an oceanographic cruise on the Ross Sea when winds arrived to the Ross Sea from the Southern Ocean, levoglucosan was present in higher concentrations in comparison to levels observed during prevailing katabatic winds.

PCs on the Antarctic plateau were mainly observed in fine particles although the coarse fraction was considerable in the 2011-2012 Dome C samples. PCs in coastal Mario Zucchelli Station particle matter were primarily observed in fine particles, unlike levoglucosan that was also present in the coarse fraction. The Antarctic samples had different levoglucosan/PC ratios from aerosols directly affected by biomass burning. PCs in coastal samples were less oxidized in comparison to those collected on the plateau, as demonstrated by low acid/aldehyde ratios at the coastal site.

451

The differences between levoglucosan and PCs, regardless of sampling time or location, suggests that PCs have sources other than biomass burning. Such a non-biomass burning source is consistent with the literature including results from Ny Alesund (Arctic) (Zangrando at al., 2013) and in Alert (Canadian Arctic) (Fu at al., 2009). The lack of vegetation in Antarctica excludes plant debris as a 456 possible source of PCs and suggests that the ocean is a likely source, especially as lignin is present in 457 marine environments. Lignin compounds account for one third of terrestrial plant biomass, are produced in vascular plant tissues (Li at al., 2012), and are present in soils (Cecchi at al., 2004), 458 459 natural waters (Louchouarn at al., 2000), rivers (Benner and Opsahl 2001; Lobbes at al., 2000; Onstad 460 at al., 2000), in oceans both as dissolved organic matter and particulate organic matter(Hedges at al., 1997; Opsahl and Benner 1997; Opsahl and Benner 1998) and aerosols (Shakya at al., 2011). Lignin 461 contains methoxyphenols (Opsahl and Benner 1997) and degrades photochemically (Opsahl and 462 463 Benner 1998) and microbiologically (Edelkraut 1996). Although the degradation of lignin in water is 464 described in the literature, only a limited number of publications determine free PCs in water (Opsahl 465 and Benner 1995), where these studies are limited to free vanillin in rivers and sounds (Edelkraut 466 1996; Keil at al., 2011). Other oceanic sources of PCs include the aging of Emiliana Huxlevi 467 (Seyedsayamdost at al., 2011), an ubiquitous marine microalga present in tropical to polar waters that 468 produces PA as an algal lignin breakdown product, which may then be transferred to the atmosphere. 469 In addition, the relatively high percentage of HA at the Mario Zucchelli Station coastal site (29% of 470 the coarse fraction) and in marine aerosols collected during the oceanographic cruise indicates that 471 this compound is a possible marine biomarker.

472

We determined that PCs are reliable tracers close to their source, yet their atmospheric concentrations may be influenced by other sources after long-range transport. Further studies on the geographical variations of PCs at their source and after transport are essential in order to fully understand their applicability as environmental markers

477

#### 478 Acknowledgments

This work was financially supported by the Italian Programma Nazionale di Ricerche in Antartide
(PNRA) through the project "Studio delle sorgenti e dei processi di trasferimento dell'aerosol
atmosferico antartico" (2009/A2.11) (contribution n° x).

| 482 | The research was also supported by funding from the National Research Council of Italy (CNR) and |
|-----|--|
| 483 | from the Early Human Impact ERC Advanced Grant from the European Commission's VII                |
| 484 | Framework Programme, grant number 267696, contribution n° y.                                     |

485 The authors thank ELGA LabWater for providing the PURE-LAB Option-R and Ultra Analytic,
486 which produced the ultra-pure water used in these experiments.

487 The authors gratefully acknowledge the NOAA Air Resources Laboratory (ARL) for providing the

488 HYSPLIT transport and dispersion model and/or READY website (http://www.ready.noaa.gov) used
489 in this publication.

490 We sincerely thank Prof. A. Ceccarini (University of Pisa, Italy), Dr. M. Bonazza (University of

491 Trieste, Italy), Dr. S. Illuminati (Polytechnic University of Marche – Ancona, Italy) and Dr. E. Padoan

492 (University of Torino, Italy) for their help and cooperation during the sampling activities in493 Antarctica.

494

#### 495 Supporting Information Available

496 Navigation details and cruise tracks are available in Table S1 and Figure S10. The atmospheric

497 concentrations (Table S2) and distributions (Figures S6, and S7) of molecular tracers for all Antarctic

498 campaigns are reported. Back-trajectories, CMBT and vector wind plots are reported in Figures S1A-

499 1C, S2-S5, S8A-G, S11A-B. The NASA, FIRMS Web Fire Mapper data is reported in S10.

- 501 **References**
- 502
- Agarwal, S.; Aggarwal, S.G.; Okuzawa, K.; Kawamura, K. Size distributions of dicarboxylic acids,
   ketoacids, alpha-dicarbonyls, sugars, WSOC, OC, EC and inorganic ions in atmospheric
   particles over Northern Japan: implication for long-range transport of Siberian biomass
   burning and East Asian polluted aerosols. Atmospheric Chemistry and Physics. 10:5839 5858; 2010 DOI: 10.5194/acp-10-5839-2010
- Akagi, S.K.; Yokelson, R.J.; Wiedinmyer, C.; Alvarado, M.J.; Reid, J.S.; Karl, T.; Crounse, J.D.;
  Wennberg, P.O. Emission factors for open and domestic biomass burning for use in atmospheric models. Atmospheric Chemistry and Physics. 11:4039-4072; 2011 doi:10.5194/acp-11-4039-2011

- Argentini, S.; Viola, A.; Sempreviva, A.; Petenko, I. Summer boundary-layer height at the plateau
   site of Dome C, Antarctica. Boundary-Layer Meteorology. 115:409-422; 2005 DOI
   10.1007/s10546-004-5643-6
- Asmi, E.; Frey, A.; Virkkula, A.; Ehn, M.; Manninen, H.E.; Timonen, H.; Tolonen-Kivimäki, O.;
  Aurela, M.; Hillamo, R.; Kulmala, M. Hygroscopicity and chemical composition of
  Antarctic sub-micrometre aerosol particles and observations of new particle formation.
  Atmos Chem Phys. 10:4253-4271; 2010 DOI:10.5194/acp-10-4253-2010
- Ault, A.P.; Moffet, R.C.; Baltrusaitis, J.; Collins, D.B.; Ruppel, M.J.; Cuadra-Rodriguez, L.A.;
  Zhao, D.; Guasco, T.L.; Ebben, C.J.; Geiger, F.M.; Bertram, T.H.; Prather, K.A.; Grassian,
  V.H. Size-Dependent Changes in Sea Spray Aerosol Composition and Properties with
  Different Seawater Conditions. Environmental Science & Technology. 47:5603-5612; 2013
  DOI: 10.1021/es400416g
- Bari, M.A.; Baumbach, G.; Kuch, B.; Scheffknecht, G. Wood smoke as a source of particle-phase
   organic compounds in residential areas. Atmospheric Environment. 43:4722-4732; 2009
   DOI: 10.4209/aaqr.2010.09.0079
- Bari, M.A.; Baumbach, G.; Kuch, B.; Scheffknecht, G. Temporal variation and impact of wood
   smoke pollution on a residential area in southern Germany. Atmospheric Environment.
   44:3823-3832; 2010 DOI:10.1016/j.atmosenv.2010.06.031
- Bari, M.A.; Baumbach, G.; Kuch, B.; Scheffknecht, G. Air Pollution in Residential Areas from
  Wood-fired Heating. Aerosol and Air Quality Research. 11:749-757; 2011 DOI:
  10.4209/aaqr.2010.09.0079
- Benner, R.; Opsahl, S. Molecular indicators of the sources and transformations of dissolved organic
   matter in the Mississippi river plume. Organic Geochemistry. 32:597-611; 2001
   DOI:10.1016/S0146-6380(00)00197-2
- Cecchi, A.M.; Koskinen, W.C.; Cheng, H.H.; Haider, K. Sorption-desorption of phenolic acids as
   affected by soil properties. Biology and Fertility of Soils. 39:235-242; 2004 DOI
   10.1007/s00374-003-0710-6
- Cheng, Y.; Engling, G.; He, K.B.; Duan, F.K.; Ma, Y.L.; Du, Z.Y.; Liu, J.M.; Zheng, M.; Weber,
  R.J. Biomass burning contribution to Beijing aerosol. Atmospheric Chemistry and Physics.
  13:7765-7781; 2013 doi:10.5194/acp-13-7765-2013
- 542 Coeur-Tourneur, C.; Cassez, A.; Wenger, J.C. Rate Coefficients for the Gas-Phase Reaction of
   543 Hydroxyl Radicals with 2-Methoxyphenol (Guaiacol) and Related Compounds. Journal of
   544 Physical Chemistry A. 114:11645-11650; 2010 DOI: 10.1021/jp1071023
- 545 Draxler, R.R.; Rolph, G.D. HYSPLIT (HYbrid Single-Particle Lagrangian Integrated Trajectory)
   546 Model access via NOAA ARL READY Website (http://www.arl.noaa.gov/HYSPLIT.php).
   547 NOAA Air Resources Laboratory, College Park, MD; 2013
- 548 Dutton, S.J.; Rajagopalan, B.; Vedal, S.; Hannigan, M.P. Temporal patterns in daily measurements
   549 of inorganic and organic speciated PM(2.5) in Denver. Atmospheric Environment. 44:987 550 998; 2010 DOI:10.1016/j.atmosenv.2009.06.006
- Dutton, S.J.; Williams, D.E.; Garcia, J.K.; Vedal, S.; Hannigan, M.P. PM(2.5) characterization for
   time series studies: Organic molecular marker speciation methods and observations from
   daily measurements in Denver. Atmospheric Environment. 43:2018-2030; 2009
   DOI:10.1016/j.atmosenv.2009.01.003
- Edelkraut, F. Dissolved vanillin as tracer for estuarine lignin conversion. Estuarine Coastal and
   Shelf Science. 43:737-745; 1996 DOI:10.1006/ecss.1996.0100
- Fattori, I.; Becagli, S.; Bellandi, S.; Castellano, E.; Innocenti, M.; Mannini, A.; Severi, M.; Vitale,
  V.; Udisti, R. Chemical composition and physical features of summer aerosol at Terra Nova
  Bay and Dome C, Antarctica. Journal of Environmental Monitoring. 7:1265-1274; 2005
  DOI: 10.1039/B507327H

- Fiebig, M.; Lunder, C.R.; Stohl, A. Tracing biomass burning aerosol from South America to Troll
   Research Station, Antarctica. Geophysical Research Letters. 36:5; 2009
   DOI:10.1029/2009GL038531
- Flanner, M.G.; Zender, C.S.; Randerson, J.T.; Rasch, P.J. Present-day climate forcing and response
  from black carbon in snow. Journal of Geophysical Research-Atmospheres. 112; 2007
  DOI:10.1029/2006JD008003
- Fu, P.Q.; Kawamura, K.; Barrie, L.A. Photochemical and Other Sources of Organic Compounds in
   the Canadian High Arctic Aerosol Pollution during Winter-Spring. Environmental Science
   & Technology. 43:286-292; 2009 DOI: 10.1021/es803046q
- Fu, P.Q.; Kawamura, K.; Pavuluri, C.M.; Swaminathan, T.; Chen, J. Molecular characterization of
   urban organic aerosol in tropical India: contributions of primary emissions and secondary
   photooxidation. Atmospheric Chemistry and Physics. 10:2663-2689; 2010 doi:10.5194/acp 10-2663-2010
- Gambaro, A.; Zangrando, R.; Gabrielli, P.; Barbante, C.; Cescon, P. Direct determination of
   levoglucosan at the picogram per milliliter level in Antarctic ice by high-performance liquid
   chromatography/electrospray ionization triple quadrupole mass spectrometry. Analytical
   Chemistry. 80:1649-1655; 2008 DOI: 10.1021/ac701655x
- Graham, B.; Mayol-Bracero, O.L.; Guyon, P.; Roberts, G.C.; Decesari, S.; Facchini, M.C.; Artaxo,
  P.; Maenhaut, W.; Koll, P.; Andreae, M.O. Water-soluble organic compounds in biomass
  burning aerosols over Amazonia 1. Characterization by NMR and GC-MS. Journal of
  Geophysical Research-Atmospheres. 107:16; 2002 DOI: 10.1029/2001JD000336
- Hara, K.; Osada, K.; Yabuki, M.; Hashida, G.; Yamanouchi, T.; Hayashi, M.; Shiobara, M.; Nishita,
  C.; Wada, M. Haze episodes at Syowa Station, coastal Antarctica: Where did they come
  from? Journal of Geophysical Research-Atmospheres. 115:20; 2010
  DOI:10.1029/2009JD012582
- Hawthorne, S.B.; Miller, D.J.; Langenfeld, J.J.; Krieger, M.S. PM-10 high-volume collection and
  quantitation of semivolatile and nonvolatile phenols, methoxylated phenols, alkanes, and
  polycyclic aromatic-hydrocarbons from winter urban air and their relationship to wood
  smoke emissions. Environmental Science & Technology. 26:2251-2262; 1992 DOI:
  10.1021/es00035a026
- He, J.; Zielinska, B.; Balasubramanian, R. Composition of semi-volatile organic compounds in the
   urban atmosphere of Singapore: influence of biomass burning. Atmospheric Chemistry and
   Physics. 10:11401-11413; 2010 DOI:10.5194/acp-10-11401-2010
- Hedges, J.I.; Keil, R.G.; Benner, R. What happens to terrestrial organic matter in the ocean?
   Organic Geochemistry. 27:195-212; 1997 DOI:10.1016/S0146-6380(97)00066-1
- Hegg, D.A.; Warren, S.G.; Grenfell, T.C.; Doherty, S.J.; Clarke, A.D. Sources of light-absorbing
  aerosol in arctic snow and their seasonal variation. Atmospheric Chemistry and Physics.
  10:10923-10938; 2010 DOI:10.5194/acp-10-10923-2010
- Hennigan, C.J.; Sullivan, A.P.; Collett, J.L.; Robinson, A.L. Levoglucosan stability in biomass
   burning particles exposed to hydroxyl radicals. Geophysical Research Letters. 37; 2010
   DOI:10.1029/2010GL043088
- Herckes, P.; Engling, G.; Kreidenweis, S.M.; Collett, J.L. Particle size distributions of organic
  aerosol constituents during the 2002 Yosemite Aerosol Characterization Study.
  Environmental Science & Technology. 40:4554-4562; 2006 DOI: 10.1021/es0515396
- Hobbs, P.V.; Reid, J.S.; Kotchenruther, R.A.; Ferek, R.J.; Weiss, R. Direct radiative forcing by
  smoke from biomass burning. Science. 275:1776-1778; 1997
  DOI:10.1126/science.275.5307.1777
- Hoffmann, D.; Tilgner, A.; Iinuma, Y.; Herrmann, H. Atmospheric Stability of Levoglucosan: A
   Detailed Laboratory and Modeling Study. Environmental Science & Technology. 44:694 610 699; 2010 DOI: 10.1021/es902476f

- Hu, Q.H.; Xie, Z.Q.; Wang, X.M.; Kang, H.; Zhang, P.F. Levoglucosan indicates high levels of
   biomass burning aerosols over oceans from the Arctic to Antarctic. Scientific Reports. 3:7;
   2013 DOI:10.1038/srep03119
- 614 Iinuma, Y.; Bruggemann, E.; Gnauk, T.; Muller, K.; Andreae, M.O.; Helas, G.; Parmar, R.;
  615 Herrmann, H. Source characterization of biomass burning particles: The combustion of
  616 selected European conifers, African hardwood, savanna grass, and German and Indonesian
- 617 peat. Journal of Geophysical Research-Atmospheres. 112; 2007
- 618 DOI:10.1029/2006JD007120
- 619 IPCC 2013: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to
  620 the Fifth Assessment Report of the Intergovernmental Panel on Climate Change. in: Stocker
  621 T.F., D. Qin, G.-K. Plattner, M. Tignor, S.K. Allen, J. Boschung, A. Nauels, Y. Xia, V. Bex
  622 and P.M. Midgley, ed. Cambridge, United Kingdom and New York, NY, USA; 2013
- Jarvinen, E.; Virkkula, A.; Nieminen, T.; Aalto, P.P.; Asmi, E.; Lanconelli, C.; Busetto, M.; Lupi,
- A.; Schioppo, R.; Vitale, V.; Mazzola, M.; Petaja, T.; Kerminen, V.M.; Kulmala, M.
  Seasonal cycle and modal structure of particle number size distribution at Dome C,
  Antarctica. Atmospheric Chemistry and Physics. 13:7473-7487; 2013 DOI:10.5194/acp-137473-2013
- Kawamura, K.; Izawa, Y.; Mochida, M.; Shiraiwa, T. Ice core records of biomass burning tracers
  (levoglucosan and dehydroabietic, vanillic and p-hydroxybenzoic acids) and total organic
  carbon for past 300 years in the Kamchatka Peninsula, Northeast Asia. Geochimica Et
  Cosmochimica Acta. 99:317-329; 2012 DOI:10.1016/j.gca.2012.08.006
- Kehrwald, N.; Zangrando, R.; Gabrielli, P.; Jaffrezo, J.L.; Boutron, C.; Barbante, C.; Gambaro, A.
  Levoglucosan as a specific marker of fire events in Greenland snow. Tellus, Series B:
  Chemical and Physical Meteorology. 64; 2012 DOI
  http://dx.doi.org/10.2402/tellush.v64i0.18106
- 635 http://dx.doi.org/10.3402/tellusb.v64i0.18196
- Keil, R.; Salemme, K.; Forrest, B.; Neibauer, J.; Logsdon, M. Differential presence of
  anthropogenic compounds dissolved in the marine waters of Puget Sound, WA and Barkley
  Sound, BC. Marine Pollution Bulletin. 62:2404-2411; 2011
  DOI:10.1016/j.marpolbul.2011.08.029
- Kessler, S.H.; Smith, J.D.; Che, D.L.; Worsnop, D.R.; Wilson, K.R.; Kroll, J.H. Chemical Sinks of
  Organic Aerosol: Kinetics and Products of the Heterogeneous Oxidation of Erythritol and
  Levoglucosan. Environmental Science & Technology. 44:7005-7010; 2010 DOI:
  10.1021/es101465m
- King J.C.; Turner J., 1997*Antarctic Meteorology and Clymatology*. Cambridge University Press:
   New York (USA).
- Kleeman, M.J.; Riddle, S.G.; Jakober, C.A. Size distribution of particle-phase molecular markers
  during a severe winter pollution episode. Environmental Science & Technology. 42:64696475; 2008 DOI: 10.1021/es800346k
- Kundu, S.; Kawamura, K.; Andreae, T.W.; Hoffer, A.; Andreae, M.O. Diurnal variation in the
  water-soluble inorganic ions, organic carbon and isotopic compositions of total carbon and
  nitrogen in biomass burning aerosols from the LBA-SMOCC campaign in Rondonia, Brazil.
  Journal of Aerosol Science. 41:118-133; 2010 DOI:10.1016/j.jaerosci.2009.08.006
- Kuo, L.J.; Louchouarn, P.; Herbert, B.E. Influence of combustion conditions on yields of solvent extractable anhydrosugars and lignin phenols in chars: Implications for characterizations of
   biomass combustion residues. Chemosphere. 85:797-805; 2011
   DOI:10.1016/i.chemosphere.2011.06.074
- 656 DOI:10.1016/j.chemosphere.2011.06.074
- Legrand, M.; Preunkert, S.; Schock, M.; Cerqueira, M.; Kasper-Giebl, A.; Afonso, J.; Pio, C.;
  Gelencser, A.; Dombrowski-Etchevers, I. Major 20th century changes of carbonaceous
  aerosol components (EC, WinOC, DOC, HULIS, carboxylic acids, and cellulose) derived
  from Alpine ice cores. Journal of Geophysical Research-Atmospheres. 112:12; 2007 DOI:
  10.1029/2006JD008080

- Li, X.G.; Zhang, T.; Sun, S.W.; Lan, H.Q.; Yu, T. Lignin in marine environment and its analysis-A
  review. Journal of Ocean University of China. 11:501-506; 2012 DOI 10.1007/s11802-0121834-9 10.1007/s11802
- Li, Y.J.; Huang, D.D.; Cheung, H.Y.; Lee, A.K.Y.; Chan, C.K. Aqueous-phase photochemical
  oxidation and direct photolysis of vanillin a model compound of methoxy phenols from
  biomass burning. Atmos Chem Phys. 14:2871-2885; 2014 DOI:10.5194/acp-14-2871-2014
- Liu, C.G.; Zhang, P.; Wang, Y.F.; Yang, B.; Shu, J.N. Heterogeneous Reactions of Particulate
   Methoxyphenols with NO3 Radicals: Kinetics, Products, and Mechanisms. Environmental
   Science & Technology. 46:13262-13269; 2012 DOI 10.1021/es303889z
- Lobbes, J.M.; Fitznar, H.P.; Kattner, G. Biogeochemical characteristics of dissolved and particulate
   organic matter in Russian rivers entering the Arctic Ocean. Geochimica Et Cosmochimica
   Acta. 64:2973-2983; 2000 DOI:10.1016/S0016-7037(00)00409-9
- Louchouarn, P.; Opsahl, S.; Benner, R. Isolation and quantification of dissolved lignin from natural
   waters using solid-phase extraction and GC/MS. Analytical Chemistry. 72:2780-2787; 2000
   DOI: 10.1021/ac9912552
- Mavrocordatos, D.; Kaegi, R.; Schmatloch, V. Fractal analysis of wood combustion aggregates by
   contact mode atomic force microscopy. Atmospheric Environment. 36:5653-5660; 2002
   DOI:10.1016/S1352-2310(02)00702-1
- McConnell, J.R.; Edwards, R.; Kok, G.L.; Flanner, M.G.; Zender, C.S.; Saltzman, E.S.; Banta, J.R.;
   Pasteris, D.R.; Carter, M.M.; Kahl, J.D.W. 20th-century industrial black carbon emissions
   altered arctic climate forcing. Science. 317:1381-1384; 2007 DOI: 10.1126/science.1144856
- Mochida, M.; Kawamura, K. Hygroscopic properties of levoglucosan and related organic
   compounds characteristic to biomass burning aerosol particles. Journal of Geophysical
   Research-Atmospheres. 109:8; 2004 DOI: 10.1029/2004JD004962
- Murphey, B.B.; Hogan, A.W. Meteorological transport of continental soot to Antarctica.
   Geophysical Research Letters. 19:33-36; 1992 DOI:10.1029/91GL02912
- Net, S.; Alvarez, E.G.; Gligorovski, S.; Wortham, H. Heterogeneous reactions of ozone with
   methoxyphenols, in presence and absence of light. Atmospheric Environment. 45:3007 3014; 2011 DOI:10.1016/j.atmosenv.2011.03.026
- 691 NOAA. ARL, Air Resource Laboratory, http://www.arl.noaa.gov/faq\_hg14.php.
- Novakov, T.; Corrigan, C.E. Cloud condensation nucleus activity of the organic component of
   biomass smoke particles. Geophysical Research Letters. 23:2141-2144; 1996
   DOI:10.1029/96GL01971.
- 695 O'Dowd, C.D.; De Leeuw, G. Marine aerosol production: a review of the current knowledge.
   696 Philosophical Transactions of the Royal Society a-Mathematical Physical and Engineering
   697 Sciences. 365:1753-1774; 2007 DOI: 10.1098/rsta.2007.2043
- O'Dowd, C.D.; Facchini, M.C.; Cavalli, F.; Ceburnis, D.; Mircea, M.; Decesari, S.; Fuzzi, S.; Yoon,
   Y.J.; Putaud, J.P. Biogenically driven organic contribution to marine aerosol. Nature.
   431:676-680; 2004 DOI:10.1038/nature02959
- Onstad, G.D.; Canfield, D.E.; Quay, P.D.; Hedges, J.I. Sources of particulate organic matter in rivers from the continental USA: Lignin phenol and stable carbon isotope compositions.
   Geochimica Et Cosmochimica Acta. 64:3539-3546; 2000 DOI: 10.1016/s0016-7037(00)00451-8
- Opsahl, S.; Benner, R. Early diagenesis of vascular plant-tissues-lignin and cutin decomposition
   and biogechemical implications. Geochimica Et Cosmochimica Acta. 59:4889-4904; 1995
   DOI:10.1016/0016-7037(95)00348-7
- Opsahl, S.; Benner, R. Distribution and cycling of terrigenous dissolved organic matter in the
   ocean. Nature. 386:480-482; 1997 DOI:10.1038/386480a0
- Opsahl, S.; Benner, R. Photochemical reactivity of dissolved lignin in river and ocean waters.
   Limnology and Oceanography. 43:1297-1304; 1998 DOI: 10.4319/lo.1998.43.6.1297

- Oros, D.R.; bin Abas, M.R.; Omar, N.; Rahman, N.A.; Simoneit, B.R.T. Identification and emission
   factors of molecular tracers in organic aerosols from biomass burning: Part 3. Grasses.
   Applied Geochemistry. 21:919-940; 2006 DOI:10.1016/j.apgeochem.2006.01.008
- Oros, D.R.; Simoneit, B.R.T. Identification and emission factors of molecular tracers in organic
   aerosols from biomass burning Part 1. Temperate climate conifers. Applied Geochemistry.
   16:1513-1544; 2001a DOI:10.1016/S0883-2927(01)00021-X
- Oros, D.R.; Simoneit, B.R.T. Identification and emission factors of molecular tracers in organic
   aerosols from biomass burning Part 2. Deciduous trees. Applied Geochemistry. 16:1545 1565; 2001b DOI:10.1016/S0883-2927(01)00022-1
- Pant, V.; Siingh, D.; Kamra, A.K. Size distribution of atmospheric aerosols at Maitri, Antarctica.
   Atmospheric Environment. 45:5138-5149; 2011 DOI:10.1016/j.atmosenv.2011.06.028
- Pereira, E.B.; Evangelista, H.; Pereira, K.C.D.; Cavalcanti, I.F.A.; Setzer, A.W. Apportionment of
   black carbon in the South Shetland Islands, Antarctic Peninsula. Journal of Geophysical
   Research-Atmospheres. 111:14; 2006 DOI:10.1029/2005JD006086
- Perrone, M.G.; Larsen, B.R.; Ferrero, L.; Sangiorgi, G.; De Gennaro, G.; Udisti, R.; Zangrando, R.;
  Gambaro, A.; Bolzacchini, E. Sources of high PM2.5 concentrations in Milan, Northern
  Italy: Molecular marker data and CMB modelling. Science of the Total Environment.
  414:343-355; 2012 DOI:10.1016/j.scitotenv.2011.11.026
- Petzold A., Karcher B. 2012, Chapter 3, In Schumann U. (Ed), Atmospheric Physics Background—
   Methods—Trends. Heidelberg: Springer-Verlag Berlin. DOI 10.1007/978-3-642-30183-4
- PNRA. Progetto Nazionale di Ricerca in Antartide, Rapporti di campagna.
   http://www.pnra.it/biblioteca/rapporti campagna.htm
- Raes, F.; Van Dingenen, R.; Vignati, E.; Wilson, J.; Putaud, J.P.; Seinfeld, J.H.; Adams, P.
  Formation and cycling of aerosols in the global troposphere. Atmospheric Environment.
  34:4215-4240; 2000 DOI:10.1016/S1352-2310(00)00239-9
- Ramanathan, V.; Carmichael, G. Global and regional climate changes due to black carbon. Nature
   Geoscience. 1:221-227; 2008 DOI:10.1038/ngeo156
- Roiger A.; Huntrieser H.; Schlager H. Chapter 12, In Schumann U. (Ed), Atmospheric Physics
  Background—Methods—Trends. Heidelberg: Springer-Verlag Berlin. DOI 10.1007/978-3642-30183-4
- Saiz-Lopez, A.; Plane, J.M.C.; Mahajan, A.S.; Anderson, P.S.; Bauguitte, S.J.B.; Jones, A.E.;
  Roscoe, H.K.; Salmon, R.A.; Bloss, W.J.; Lee, J.D.; Heard, D.E. On the vertical distribution
  of boundary layer halogens over coastal Antarctica: implications for O-3, HOx, NOx and the
  Hg lifetime. Atmospheric Chemistry and Physics. 8:887-900; 2008
- Schkolnik, G.; Falkovich, A.H.; Rudich, Y.; Maenhaut, W.; Artaxo, P. New analytical method for
  the determination of levoglucosan, polyhydroxy compounds, and 2-methylerythritol and its
  application to smoke and rainwater samples. Environmental Science & Technology.
  39:2744-2752; 2005 DOI: 10.1021/es048363c
- Schmitt-Kopplin, P.; Liger-Belair, G.; Koch, B.P.; Flerus, R.; Kattner, G.; Harir, M.; Kanawati, B.;
  Lucio, M.; Tziotis, D.; Hertkorn, N.; Gebefügi, I. Dissolved organic matter in sea spray: a
  transfer study from marine surface water to aerosols. Biogeosciences. 9:1571-1582; 2012
  DOI:10.5194/bgd-8-11767-2011
- Seyedsayamdost, M.R.; Case, R.J.; Kolter, R.; Clardy, J. The Jekyll-and-Hyde chemistry of
   Phaeobacter gallaeciensis. Nature Chemistry. 3:331-335; 2011 DOI:10.1038/nchem.1002
- Shakya, K.M.; Louchouarn, P.; Griffin, R.J. Lignin-Derived Phenols in Houston Aerosols:
   Implications for Natural Background Sources. Environmental Science & Technology.
   45:8268-8275; 2011 DOI: 10.1021/es201668y
- Simoneit, B.R.T. Biomass burning A review of organic tracers for smoke from incomplete
   combustion. Applied Geochemistry. 17:129-162; 2002 DOI:10.1016/S0883-2927(01)00061 0

762 Simpson, C.D.; Dills, R.L.; Katz, B.S.; Kalman, D.A. Determination of levoglucosan in atmospheric fine particulate matter. Journal of the Air & Waste Management Association. 763 764 54:689-694; 2004 DOI:10.1080/10473289.2004.10470945 765 Simpson, C.D.; Paulsen, M.; Dills, R.L.; Liu, L.J.S.; Kalman, D.A. Determination of methoxyphenols in ambient atmospheric particulate matter: Tracers for wood combustion. 766 767 Environmental Science & Technology. 39:631-637; 2005 DOI: 10.1021/es0486871 768 Smith, J.D.; Sio, V.; Yu, L.; Zhang, Q.; Anastasio, C. Secondary Organic Aerosol Production from Aqueous Reactions of Atmospheric Phenols with an Organic Triplet Excited State. 769 770 Environmental Science & Technology. 48:1049-1057; 2014 DOI: 10.1021/es4045715 771 Stohl, A.; Andrews, E.; Burkhart, J.F.; Forster, C.; Herber, A.; Hoch, S.W.; Kowal, D.; Lunder, C.; 772 Mefford, T.; Ogren, J.A.; Sharma, S.; Spichtinger, N.; Stebel, K.; Stone, R.; Strom, J.; 773 Torseth, K.; Wehrli, C.; Yttri, K.E. Pan-Arctic enhancements of light absorbing aerosol concentrations due to North American boreal forest fires during summer 2004. Journal of 774 775 Geophysical Research-Atmospheres. 111; 2006 DOI: 10.1029/2006jd007216 776 Stohl, A.; Berg, T.; Burkhart, J.F.; Fjaeraa, A.M.; Forster, C.; Herber, A.; Hov, O.; Lunder, C.; McMillan, W.W.; Oltmans, S.; Shiobara, M.; Simpson, D.; Solberg, S.; Stebel, K.; Strom, 777 J.; Torseth, K.; Treffeisen, R.; Virkkunen, K.; Yttri, K.E. Arctic smoke - record high air 778 779 pollution levels in the European Arctic due to agricultural fires in Eastern Europe in spring 2006. Atmospheric Chemistry and Physics. 7:511-534; 2007 DOI:10.5194/acp-7-511-2007 780 781 Stohl, A.; Sodemann, H. Characteristics of atmospheric transport into the Antarctic troposphere. Journal of Geophysical Research-Atmospheres. 115; 2010 DOI:10.1029/2009JD012536 782 783 Svenningsson, B.; Rissler, J.; Swietlicki, E.; Mircea, M.; Bilde, M.; Facchini, M.C.; Decesari, S.; 784 Fuzzi, S.; Zhou, J.; Mønster, J.; Rosenørn, T. Hygroscopic growth and critical 785 supersaturations for mixed aerosol particles of inorganic and organic compounds of 786 atmospheric relevance. Atmos Chem Phys. 6:1937-1952; 2006 DOI:10.5194/acp-6-1937-2006 787 788 Taylor, D. Biomass burning, humans and climate change in Southeast Asia. Biodiversity and 789 Conservation. 19:1025-1042; 2010 DOI 10.1007/s10531-009-9756-6 790 Udisti, R.; Dayan, U.; Becagli, S.; Busetto, M.; Frosini, D.; Legrand, M.; Lucarelli, F.; Preunkert, 791 S.; Severi, M.; Traversi, R.; Vitale, V. Sea spray aerosol in central Antarctica. Present 792 atmospheric behaviour and implications for paleoclimatic reconstructions. Atmospheric 793 Environment. 52:109-120; 2012 DOI:10.1016/j.atmosenv.2011.10.018 794 Vestin, A.; Rissler, J.; Swietlicki, E.; Frank, G.P.; Andreae, M.O. Cloud-nucleating properties of the 795 Amazonian biomass burning aerosol: Cloud condensation nuclei measurements and 796 modeling. Journal of Geophysical Research: Atmospheres. 112:D14201; 2007 797 DOI:10.1029/2006JD008104 798 von Schneidemesser, E.; Schauer, J.J.; Hagler, G.S.W.; Bergin, M.H. Concentrations and sources of 799 carbonaceous aerosol in the atmosphere of Summit, Greenland. Atmospheric Environment. 800 43:4155-4162; 2009 DOI:10.1016/j.atmosenv.2009.05.043 801 Wang, G.H.; Chen, C.L.; Li, J.J.; Zhou, B.H.; Xie, M.J.; Hu, S.Y.; Kawamura, K.; Chen, Y. 802 Molecular composition and size distribution of sugars, sugar-alcohols and carboxylic acids 803 in airborne particles during a severe urban haze event caused by wheat straw burning. 804 Atmospheric Environment. 45:2473-2479; 2011 DOI:10.1016/j.atmosenv.2011.02.045 805 Wang, G.H.; Kawamura, K.; Lee, M. Comparison of organic compositions in dust storm and normal aerosol samples collected at Gosan, Jeju Island, during spring 2005. Atmospheric 806 807 Environment. 43:219-227; 2009 DOI:10.1016/j.atmosenv.2011.02.045 808 Ward, T.J.; Hamilton, R.F.; Dixon, R.W.; Paulsen, M.; Simpson, C.D. Characterization and 809 evaluation of smoke tracers in PM: Results from the 2003 Montana wildfire season. Atmospheric Environment. 40:7005-7017; 2006 DOI:10.1016/j.atmosenv.2006.06.034 810 811 Ward, T.J.; Palmer, C.P.; Bergauff, M.; Jayanty, R.K.M.; Noonan, C.W. Organic/elemental carbon 812 and woodsmoke tracer concentrations following a community wide woodstove changeout

- 813 program. Atmospheric Environment. 45:5554-5560; 2011
- 814 DOI:10.1016/j.atmosenv.2011.05.005
- Wolff, E.W.; Cachier, H. Concentrations and seasonal cycle of black carbon in aerosol at a coastal
   Antarctic station. Journal of Geophysical Research-Atmospheres. 103:11033-11041; 1998
   DOI:10.1029/97JD01363
- Yao, P.; Schwab, V.F.; Roth, V.N.; Xu, B.Q.; Yao, T.D.; Gleixner, G. Levoglucosan concentrations
  in ice-core samples from the Tibetan Plateau determined by reverse-phase high-performance
  liquid chromatography mass spectrometry. Journal of Glaciology. 59:599-612; 2013 DOI:
  http://dx.doi.org/10.3189/2013JoG12J157
- Yttri, K.E.; Lund Myhre, C.; Eckhardt, S.; Fiebig, M.; Dye, C.; Hirdman, D.; Ström, J.; Klimont, Z.;
  Stohl, A. Quantifying black carbon from biomass burning by means of levoglucosan a
  one-year time series at the Arctic observatory Zeppelin. Atmos Chem Phys. 14:6427-6442;
  2014 DOI:10.5194/acpd-13-31965-2013
- Yu, L.; Smith, J.; Laskin, A.; Anastasio, C.; Laskin, J.; Zhang, Q. Chemical characterization of
  SOA formed from aqueous-phase reactions of phenols with the triplet excited state of
  carbonyl and hydroxyl radical. Atmos Chem Phys Discuss. 14:21149-21187; 2014
  DOI:10.5194/acpd-14-21149-2014
- Zangrando, R.; Barbaro, E.; Zennaro, P.; Rossi, S.; Kehrwald, N.M.; Gabrieli, J.; Barbante, C.;
  Gambaro, A. Molecular Markers of Biomass Burning in Arctic Aerosols. Environmental
  Science & Technology. 47:8565-8574; 2013 DOI: 10.1021/es400125r
- Zennaro, P.; Kehrwald, N.; McConnell, J.R.; Schüpbach, S.; Maselli, O.J.; Marlon, J.; Vallelonga,
  P.; Leuenberger, D.; Zangrando, R.; Spolaor, A.; Borrotti, M.; Barbaro, E.; Gambaro, A.;
  Barbante, C. Fire in ice: two millennia of boreal forest fire history from the Greenland
  NEEM ice core. Clim Past. 10:1905-1924; 2014 DOI:10.5194/cp-10-1905-2014
- 837

839

840

|                         | Category                              | Sampling period                     | VA/VAN   | SyA/SyAH  | LG/VA                                       | LG/VAN                                       | LG/SyA                                     | LG/SyAH                                     | Ref                         |
|-------------------------|---------------------------------------|-------------------------------------|--|---|---|--|--|---|-----------------------------|
| Missula (MT,<br>USA)    | Smoke                                 | Wild fire                           |  |   | 343   |  | 1568                                       |   | (Ward at al., 2006)         |
| Libby<br>(MT, USA)      | Urban Aerosol<br>(PM2.5)              | Winter 2003-2004                    |  |   | 51.3  |  | 2031.7                                     | 490.4                                       | (Ward at al., 2011)         |
| Libby<br>(MT, USA)      | Urban Aerosol<br>(PM2.5)              | Winter 2008-2009                    |  |   | 30.1  |  | 4117.2                                     |   | (Ward at al., 2011)         |
| NanJing<br>(China)      | Size segregate urban<br>Aerosol (PM9) | Biomass burning<br>Haze             |  |   |   | 252 (Haze) 408, 246<br>(none-Haze)           |  |   | (Ward at al., 2011)         |
| Rodônia<br>(Brazil)     | Forest Aerosol<br>(PM2.5)             | Biomass burning period              | 8.2 (day)<br>4.9 (night)                                       | 1.9 (day)<br>1.0 (night)  | 75.9 (day)<br>66.0 (night)                  | 622.0 (day)<br>324.7 (night)                 | 64.2 (day)<br>46.5(night)                  | 123.8(day)<br>47.2 (night)                  | (Kundu at al.,<br>2010)     |
| Rodônia<br>(Brazil)     | Forest Aerosol                        | Biomass burning<br>period           | <ul><li>6.9 (Pasture site)</li><li>6.2 (Forest site)</li></ul> | <ol> <li>1.5 (Pasture site)</li> <li>5.5 (Forest site)</li> </ol> | 115.4 (Pasture site)<br>344.6 (Forest site) | 796.9 (Pasture site)<br>2125.0 (Forest site) | 65.9 (Pasture site)<br>149.1 (Forest site) | 101.2 (Pasture site)<br>822.6 (Forest site) | (Graham at al., 2002)       |
| Jeju Island<br>(Korea)  | Aerosol (TSP)                         | spring                              |  |   | 134.4                                       |  | 363  |   | (Wang at al.,<br>2009)      |
| Dittausen<br>(Germany)  | Urban Aerosol<br>(PM10)               | Nov 2005-Mar 2006                   |  |   |   | 2441.0                                       |  | 53.0  | (Bari at al., 2009)         |
| Chennai (India)         | Urban Aerosol<br>(PM10)               | Winter                              | 1.3  |   | 112.0                                       | 83.0   | 54.1                                       |   | (Fu at al., 2010)           |
| Chennai (India)         | Urban Aerosol<br>(PM10)               | Summer                              | 0.9  |   | 140.5                                       | 198.2  | 179.0                                      |   | (Fu at al., 2010)           |
| Seattle<br>(WA, USA)    | Urban Aerosol<br>(PM10, PM2.5)        | Winter                              |  |   |   | 122.5 (PM10)<br>135.5 (PM2.5)                |  | 185.1 (PM10)<br>228.9 (PM2.5)               | (Simpson at al., 2004)      |
| NyAlesund<br>(Svalbard) | Size segregate<br>Aerosol (PM10)      | 25 Apr-14 Sep<br>Boreal summer      |  | 5.0   | 4.3   |  | 71.2                                       | 338.4                                       | (Zangrando at<br>al., 2013) |
| MZS                     | Aerosol size<br>sequestred (PM10)     | 29 Nov- 18 Jan<br>Austral summer    | 0.1  | 0.5   | 10.1  | 1.0  | 162.3                                      | 69.7  | this work                   |
| DC 2011-2012            | Size segregate<br>Aerosol (PM10)      | Dic 2011-Jan 2012<br>Austral summer | 0.4  | 1.3   | 3.9   | 1.1  | 10.3                                       | 12.9  | this work                   |
| DC 2012-2013            | Size segregate<br>Aerosol (PM10)      | Dic 2012-Jan 2013<br>Austral summer | 0.5  | 1.3   | 1.7   | 1.0  | 14.3                                       | 18.2  | this work                   |

Table 1: LG/VA, LG/VAN, LG/SyA and LG/SyAH ratios calculated using means reported in the literature and in this work, as well as VA/VAN
 and SyA/SyAH, ratios which indicate the oxidation of PCs. LG is the abbreviation for levoglucosan.